

Review

# Rhodium-Catalyzed Asymmetric Arylation

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ABSTRACT: Rhodium-catalyzed asymmetric arylation (RCAA) reactions provide one of the most straightforward and powerful ways to introduce aryl fragments in an enantioselective manner. The discovery of novel chiral ligands and catalytic systems is a major focus in generating optical chiralities for RCAA reactions. In the past decade, the chelating functionalities in ligands have been significantly



expanded from traditional phosphorus to interesting diene, bissulfoxide, and their hybrids. Herein we highlight the research on these distinct families of chiral ligands and describe their applications in the RCAA of arylmetals to activated alkenes, aldehydes, ketones and imines, and RCAA-tandem reactions.

KEYWORDS: rhodium-catalyzed asymmetric arylation, conjugate addition, phosphorus ligands, diene ligands, bissulfoxide ligands, hybrid ligands, activated alkenes, imines, tandem reactions

# 1. INTRODUCTION

Over the past two decades, there has been dramatic growth in using transition-metal-catalyzed reactions for important organic transformations. Notably, significant attention has been paid to apply rhodium catalysts in the formation of C−C bonds, because of their special reactivity and selectivity compared to other catalytic systems, and their friendly reaction conditions allowing in many cases to have water as a cosolvent. In principle, any reaction that involves Rh-catalyzed asymmetric addition of an aryl group to a double or triple bond is called "Rh-catalyzed asymmetric arylation" (RCAA). A typical RCAA reaction of an aryl-metallic reagent Ar-M (II) to an electrondeficient acceptor (I) proceeds as follows: First, the transmetalation of an aryl group of Ar-M (II) with rhodium species A produces the reactive aryl-rhodium B. Then the double bond or triple bond in compound I coordinates with B and subsequently inserts into Rh−Ar bond to form the adduct C, which is readily protonated under aqueous conditions to generate the product III. This RCAA process is also named as "Rh-catalyzed asymmetric hydroarylation". On the other hand, the adduct C can be trapped by other electrophiles to yield product IV. Such a transformation is often called RCAAtandem reaction (Scheme 1).

### Scheme 1. Concept of RCAA





Prior to our compilation, several authoritative reviews partially covered this subject, including contributions by Hayashi<sup>1−8</sup> and Fros<sup>9</sup> devoted to rhodium-catalyzed asymmetric conjugate addition and its related synthetic applications, and coll[e](#page-20-0)c[ti](#page-21-0)ons by Ro[vi](#page-21-0)s<sup>10</sup> and Carreira<sup>11</sup> centered upon chiral olefins as steering ligands in asymmetric catalysis. Herein we summarize the recent [de](#page-21-0)velopment [of](#page-21-0) RCAA and RCAAtandem reactions during the period 1997−2011. The review commences with the general description of RCAA, followed by three separate sections focusing on RCAA to alkenes, aldehydes/ketones, and imines, respectively. The RCAAtandem reaction is then discussed in terms of an intermolecular process with different electrophiles and various intramolecular cyclizations. For easy comparison, examples in each section are grouped by similarity, rather than by chronology.

### 2. GENERAL DESCRIPTION OF RCAA

RCAA of aryl-boronic reagents to  $\alpha$ , $\beta$ -unsaturated ketones has attracted the most attention in this research field, and is frequently chosen as a model reaction to investigate a variety of different ligand systems. More importantly, many of the characteristics discovered in the reaction with  $\alpha$ , $\beta$ -unsaturated ketones are applicable to other RCAA systems including the reactions from different alkene classes and nucleophilic arylmetallic reagents.

2.1. History Remarks. The first Rh-catalyzed arylation of arylboronic acids to enones can be traced back to a publication in 1997 by Miyaura and co-workers.<sup>12</sup> In this report, the combination of  $[Rh-(acac)(CO)_2]$  (acac = acetylacetonato) and bisphosphine ligand dppb (d[pp](#page-21-0)b = 1,4-bis- (diphenylphosphino)butane) could efficiently catalyze the conjugate addition of arylboronic acids to linear and cyclic  $\alpha$ , $\beta$ -unsaturated ketones in various aqueous cosolvent systems

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although in slightly lower yield for 2-cyclohexenone (1b) (Scheme 2).

# Scheme 2. Initial Report of Rh-Catalyzed Arylation to Enones



The merits of this reaction lie in the following: (i) neutral reaction conditions in the presence of water; (ii) no observation of either competitive uncatalyzed reaction of the arylboronic acids to the enones or the related 1,2-addition byproduct; and (iii) tolerance of a wide range of functional groups in organoboron reagents which are in contrast to organolithium and Grignard reagents.

A milestone of this asymmetric methodology appeared in 1998, when Hayashi and Miyaura described the RCAA of arylboronic acids to  $\alpha$ , $\beta$ -unsaturated ketones.<sup>13</sup> For the first time, a broad range of aryl groups could be introduced into  $\alpha$ , $\beta$ unsaturated ketones in high yields and exc[elle](#page-21-0)nt enantioselectivities using (S)-BINAP as the ligand (Scheme 3). There



were multiple modifications from the original Miyaura's conditions: (i) different Rh-precursor, from  $[Rh(acac)(CO)<sub>2</sub>]$ to  $[Rh(\text{acac})(C_2H_4)_2]$ ; (ii) different solvent system, 1,4dioxane/ $H_2O$  (10:1); (iii) higher reaction temperature (100  $^{\circ}$ C) and shorter reaction time (5 h). Under the optimal reaction conditions, aryl groups with either electron-donating or electron-withdrawing substitutions were successfully added to both cyclic enones (2-cyclohexenone (1b), 2-cyclopentenone (1d) and 2-cycloheptenone (1e)), and trans-linear enones with excellent enantioselectivities (Scheme 3).

These key groundbreaking findings paved the way for an intense and hot research activity in the area of RCAA and related processes. Nowadays, the reaction of RCAA of organoboronic acids to  $\alpha$ , $\beta$ -unsaturated ketones is usually called Hayashi−Miyaura reaction.

2.2. Mechanism and Stereoselectivity. In 2002, Hayashi and co-workers demonstrated the detailed mechanism in the RCAA to  $\alpha$ , $\beta$ -unsaturated ketones.<sup>14</sup> An example of this catalytic cycle for the RCAA of phenylboronic acid to 2-cyclohexenone (1b) catalyzed [by](#page-21-0)  $Rh(I)/(S)$ -BINAP is illustrated in Scheme 4. The reaction is initiated through the

Scheme 4. Catalytic Cycle for the RCAA of Phenylboronic Acid to 2-Cyclohexenone by a [Rh]−OH Complex



transmetalation of a phenyl group from boron to hydroxorhodium A to generate the phenylrhodium B. Subsequently, 2-cyclohexenone inserts into Rh−Ph bond of B to form the  $\alpha$ xa- $\pi$ -allylrhodium C, which is unstable under protic condition and readily hydrolyzed to regenerate A and liberate the RCAA product 3ba. It is noteworthy that rhodium remains at a constant oxidation state of +1 throughout the catalytic cycle.

Subsequently, Hayashi and co-workers accomplished a detailed kinetic study and revealed that the transmetalation from boron to rhodium was the rate-determining step in the catalytic cycle. It was also found that the equilibrium between the catalytically inactive dimeric hydroxorhodium complex 4 or 5 and the active monomeric species A significantly affected the reaction rate (Scheme 5).<sup>15</sup> For example, the reaction rate

#### Scheme 5. Kinetic Study [of](#page-21-0) Rh-Catalyzed Arylation to MVK



catalyzed by  $[Rh(OH)(COD)]_2$  (4) was 20-fold faster than with  $[Rh(OH)((R)-BINAP)]$ <sub>2</sub> (5) under the same conditions in Rh-catalyzed arylation of  $PhB(OH)_2$  to MVK.<sup>16</sup> Such remarkably different catalytic activities between two Rhcomplexes resulted from faster rates in both equilibr[iu](#page-21-0)m and transmetalation steps by using  $Rh(I)/$ diene complex 4.

In 1998, Hayashi and co-workers proposed the stereochemical pathway for the preferential formation of the S product in RCAA reaction catalyzed by  $Rh(I)/(S)$ -BINAP complex with 2-cyclohexenone  $(1b)$  as an example (Scheme 6).<sup>13</sup> Owing to the highly skewed structure known for transition metal complexes coordinated with a BI[NA](#page-2-0)P ligand, $17$  (S)-BINA[P-](#page-21-0)Rh-Ph intermediate should have an open space at the lower part

<span id="page-2-0"></span>Scheme 6. Hayashi's Enantioselective Model for the RCAA to 2-Cyclohexenone Catalyzed by  $Rh(I)/(S)$ -BINAP Complex



of the vacant coordination site, with the upper being blocked by one of the phenyl rings of BINAP ligand. The alkenyl bond of 2 cyclohexenone can only coordinate rhodium with its  $\alpha S$ *i* face and undergo migratory insertion to form a stereogenic carbon center with S configuration in product 3ba.

In 2010, Hayashi and co-workers extended this stereochemical model to predict the absolute configurations of final products in Rh(I)/(S)-BINAP catalyzed asymmetric 1,4 addition reactions.<sup>8</sup> As shown in Scheme 7, aryl-rhodium

Scheme 7. Hayas[hi](#page-21-0)'s Stereocontrol Model for the RCAA to trans- or cis-Olefins



species selectively coordinates with the  $\alpha S$ i face of cyclic and cis-linear or  $\alpha$ Re face of trans-linear alkenes, generating corresponding chiral stereocenters in the products. Notably, the trans/cis geometry of the double bond in linear alkenes plays a dominating role in the stereochemical outcome. The linear olefins in this model refer to  $\alpha$ , $\beta$ -unsaturated ketones,  $\alpha$ , $\beta$ -unsaturated esters and amides, alkenylphosphonates, and nitroalkenes.

2.3. Aryl-Metallic Reagents and Rh Precatalysts. 2.3.1. Arylboron Reagents. The mechanism of transmetalation from boron to rhodium assumes that Rh−OH complex A can coordinate the highly oxophilic arylboronic acid 2 to give intermediate D as a quaternized boron anion, from which the aryl fragment is transferred to rhodium in an intramolecular fashion to generate the aryl-rhodium species B and boronic acid (Scheme 8).<sup>18,19</sup> This transmetalation occurs

Scheme 8. Proposed Mechanism for the Transmetallation of Arylboronic Acids to Rhodium



under neutral conditions, but it can be greatly accelerated by the addition of a stoichiometric base. This is attributable to the quaternization of arylboronic acid, which facilitates the rupture of B−Ar bond.18,20

Most arylboronic acids 2 are thermally stable and inert to water and oxyg[en, a](#page-21-0)nd they are widely used in RCAA reactions. It is found that trace amount of phenol existing in commercial phenylboronic acid can remarkably deactivate Rh(I)/chiral diene catalysts, $21$  especially in the case of low-catalyst loading. The phenol impurity can be easily removed by dehydration of boronic acid to [fo](#page-21-0)rm the cyclic trimeric anhydride (boroxine, 6), followed by washing with hexane (Figure 1). The pure boroxine





6 can be readily hydrolyzed back to the corresponding boronic acid 2 under basic aqueous conditions.<sup>22</sup> Interestingly, boroxines 6 have become one of the preferential reagents for RCAA, because of the convenient addition [in](#page-21-0) accurate stoichiometry and their better stability toward protodeboration than boronic acids, particularly at high temperature (ca. 100 $\degree$ C).

Pinacol boronic esters 7 react slowly in RCAA because of their sluggish hydrolysis back to the corresponding boronic acids. $23$  The additional coordination in N-methyliminodiacetic acid (MIDA) boronates 8 greatly improves their stability, resul[tin](#page-21-0)g in slow release of boronic acids from MIDA boronates to keep minimal amount of free boronic acid throughout the RCAA reaction.<sup>24</sup> Potassium aryltrifluoroborate salts 9 have become a popular source of organoboron reagents,<sup>25−28</sup> because they als[o e](#page-21-0)xhibit better stability than the corresponding boronic acids while staying reactive in RCAA.<sup>29</sup> [It is](#page-21-0) noteworthy that potassium organotrifluoroborates do not transmetallate directly to Rh(I), but rather thr[oug](#page-21-0)h the monohydroxyborate 14 (Scheme 9).30−<sup>34</sup>

#### Scheme 9. Transmetallation of Po[tas](#page-21-0)s[iu](#page-21-0)m Aryltrifluoroborate



Lithium trimethylarylborate salts 10 are very reactive in RCAA, but unstable and require the preparation in situ.<sup>23,35,36</sup>

Similar to MIDA (8), cyclic aryltriolborates 11 are also very stable in air and water, and conveniently used in RCAA because of better solubility in organic solvents than related potassium organotrifluoroborates 9. 37,38 The reactive ArB(9-BBN) derivatives 12 are often applied in RCAA in aprotic solvents to afford a stable chiral bor[on en](#page-21-0)olate in the absence of base,<sup>39</sup> which can be further trapped by other electrophiles to undergo tandem reactions (cf. Section 6). Sodium tetraarylborate sa[lts](#page-21-0) 13 are also very reactive reagents for the RCAA. For example, sodium tetraphenylborate (13a) reacts with Rh-chloride complex A to afford Rh-tetraphenylborate complex 15 under neutral conditions, which subsequently releases the phenylrhodium species **B** (Scheme 10).<sup>4</sup>



2.3.2. Other Aryl-Metallic Reagents. Besides boron reagents, other organometallics can also be used in RCAA reactions (Scheme 11). Theoretically, all organometallics in

# Scheme 11. Transmetallation of Aryl-Metallic Reagents with Rh

$$
Ar=[M] + [Rh]-X - \frac{transmetalation}{B} [Rh]-Ar + [M]-X
$$
  
A  

$$
M = Ti, Zn; B, Al, In; Si, Sn, Pb; Bietc
$$

which the metal is less electronegative than rhodium and all organometallic species of similar electronegativity but with weaker carbon−rhodium bonds are potential candidates for transmetalation reactions with rhodium.<sup>41</sup> For the sake of effective transmetalation, the arylmetallic reagents must be still stable under the hydrolysis conditions [of](#page-21-0) the Rh-enolate, or they can directly transmetallate with the Rh-enolate.<sup>39</sup>

 $A$ ryl-aluminum,<sup>42</sup> -titanium,<sup>43–46</sup> and -zinc<sup>47–50</sup> reagents are much more nucleophilic than the corresponding [ar](#page-21-0)ylboron species, enabling [t](#page-21-0)he trans[metalat](#page-21-0)ion at r[oom](#page-21-0) temperature under aprotic conditions.

Arylsilicon reagents are far less reactive than the corresponding boron, tin, titanium, and zinc derivatives. So far, arylsilanediols  ${\bf 16}_2^{51}$  dichlorodiarylsilanes  ${\bf 17}_2^{52}$  aryltriethoxysilanes  $18, ^{53}$  poly(phenylmethylsiloxane)  $19, ^{54}$  and [2-(hydroxylm[eth](#page-21-0)yl)aryl]dimethylsilanes 20<sup>55</sup> have b[ee](#page-21-0)n successively develope[d](#page-21-0) as the aryl-transfer reagents with [rh](#page-21-0)odium (Figure 2).



Figure 2. Arylsilicon reagents as aryl-transfer reagents.

Diarylindium hydroxides,<sup>56</sup> aryl-tin,<sup>57–59</sup> aryl-lead,<sup>60</sup> and triarylbismuth<sup>61,62</sup> reagents also prove to be competent aryltransfer agents even in t[he](#page-21-0) presen[ce of](#page-21-0) air and [w](#page-21-0)ater. Interestingly, [the](#page-21-0) reaction of [Rh]−OH with triarylmethanols 21, which are derived from acridinone 22, can produce different arylrhodiums B through β-aryl elimination of alkoxorhodium

intermediates E, providing a new effective aryl-transfer approach (Scheme 12).<sup>63</sup>





2.3.3. Rh-Precatalysts.  $[Rh(Cl)(C_2H_4)_2]_2$  is the favorable precatalyst because of its rapid and irreversible ligand exchange.<sup>13</sup> However, COD-based Rh precursors, such as  $[Rh(Cl)(COD)]_2$  and  $[Rh(OH)(COD)]_2$ , are usually avoided because [the](#page-21-0)y exhibit higher catalytic activity than chiral Rhphosphine complex.20,64 Cationic rhodium precatalysts also demonstrate fast ligand exchange,<sup>65</sup> and are often applied in RCAA. For example,  $\left[\text{Rh(COD)}_{2}\right]\text{BF}_{4}$ ,  $\left[\text{Rh(COD)}(\text{MeCN})_{2}\right]$ - $BF_4$ <sup>58,59,66–71</sup> and  $\left[\text{Rh(nbd)}_2\right]BF_4$ <sup>20,72</sup> are excellent Rh-precatalysts,<sup>73</sup> which allow for the replacement of KOH with  $Et_3N$ in [RCAA and](#page-21-0) thus make the react[ion m](#page-21-0)ore functional group-tolerant.<sup>[74](#page-21-0)</sup>

2.4. Ligand Systems. Developing new ligands and new reaction[s h](#page-21-0)as been always a major focus in the field of RCAA. In this section, an overview of the distinct ligands for RCAA of phenylboronic acid to 2-cyclohexenone (Hayashi−Miyaura reaction) will be presented. A variety of chiral ligands in RCAA reactions can be classified into the following families: phosphorus-, diene-, sulfoxide-, and their hybrid-structures. Comparison of different ligand motifs is usually performed by using Hayashi−Miyaura reaction as a model system (Scheme 13). It should be noted that this is just one

Scheme 13. RCAA of Phenylboronic Acid (2a) to 2- Cyclohexenone (1b) As a Model Reaction Using Different Ligands



comparison based on this model reaction, and some ligands might work better for other specific substrates.

2.4.1. Phosphorus Ligands. On the basis of the initial success of using chiral BINAP in RCAA, a large number of bidentate or monodentate phosphorus ligands have been applied in this area. These results are summarized in Figures 3, 5, 6, and 7.

<span id="page-4-0"></span>The BINAP-based ligands  $L_2$ <sup>75</sup>  $L_3$ <sup>75</sup> and water-soluble  $L_4$ <sup>76</sup> provided similar outcomes with [B](#page-21-0)IN[AP](#page-21-0) (Figure 3). Interesting[ly,](#page-21-0)



Figure 3.  $C_2$ -Symmetric bidentate phosphorus ligands.

BINOL-based bisphosphonites<sup>77</sup> L<sub>5</sub> and L<sub>6</sub> or L<sub>7a</sub> and L<sub>7b</sub> gave excellent but sometimes reverse enantioselectivity depending on the linker between two [sep](#page-21-0)arate phosphonites. Similarly, BINOL-based bisphosphoramidite  $L_8$  gave excellent results.<sup>74,78</sup> The bisphosphine ligand  $L_9^{79}$  bearing a distinct norbornane backbone delivered high selectivities; however, the cycloprop[ane](#page-21-0)based bisphosphines<sup>80</sup>  $L_{10a}$  a[nd](#page-21-0)  $L_{10b}$  just achieved the moderate selectivities. The biphenyl-based bisphosphorus ligands  $L_{11}$ ,  $L_{12a}$ ,  $L_{12b}$ ,  $L_{12c}$ ,  $L_{12d}$ ,  $L_{12g}$  $L_{12g}$  $L_{12g}$ ,  $L_{12b}$ , and  $L_{12i}$  likewise acquired excellent enantioselectivities,<sup>81−88</sup> in which (R)-F<sub>12</sub>−BIPHEP L<sub>12c</sub> and (R)-F<sub>24</sub> $-$ Synphos L<sub>[1](#page-21-0)2[i](#page-22-0)</sub> displayed a higher catalytic activity.

The catalytic activity of different bisphosphine ligands can be estimated by comparing their stretching frequencies  $(\nu_{\text{CO}})$  of carbonyl groups in the corresponding [RhCl(bisphos-phine)- (CO)] complexes by IR (Figure 4). The  $\nu_{\text{CO}}$  value of  $L_{12c}$  is



Figure 4.  $\nu_{\text{CO}}$  values of the [RhCl(bisphosphine)(CO)] complexes.

shown to be higher than those of known bisphosphines, suggesting that  $L_{12c}$  has stronger  $\pi$ -acidic character.<sup>82,83,85</sup> Such strong  $\pi$ -accepting ability can significantly accelerate the ratedetermining transmetalation step in Hayashi−Miya[ura r](#page-21-0)[ea](#page-22-0)ction. As a result,  $L_{12c}$  showed the highest catalytic activity with turnover frequency (TOF) and turnover number (TON) up to 54,000 h<sup>-1</sup> and 320,000.<sup>85</sup>

Axially chiral non- $C_2$ -symmetric bisphosphorus ligands  $L_{13a}$ <sup>89</sup>  $L_{13b}$ ,<sup>89</sup> and pol[ysty](#page-22-0)rene-supported BINAP  $L_{14}$ <sup>90</sup> gave excellent enantioselectivities similar to BINAP (Figure 5). The



Figure 5. Non- $C_2$ -symmetric bidentate phosphorus ligands.

bisphosphine ligands  $\mathrm{L_{15}}^{91}$   $\mathrm{L_{16}}^{92}$  and  $\mathrm{L_{17}}^{93}$  with planar chirality were also investigated in Hayashi−Miyaura reaction and only Re-based  $L_{15}$  gave signif[ica](#page-22-0)nt s[ele](#page-22-0)ctivitie[s.](#page-22-0)

The P-chiral bisphosphine ligands QuinoxP\*  $(L_{18})$ <sup>94</sup> tBu-BisP\*  $(L_{19})^{95}$   $L_{20}^{95}$  and  $L_{21}^{95}$  have been successfully developed and high enantioselectivity was achieved i[n](#page-22-0) Hayashi−Miyaura [r](#page-22-0)eacti[on](#page-22-0) (Figure [6\).](#page-22-0) The experimental results obtained with alkyne-type ligands  $L_{20}$  and  $L_{21}$  are comparable or even superior to those [ob](#page-5-0)tained with BINAP  $(L_1)$ , Quinox $P^*$  ( $L_{18}$ ), or tBu-Bis $P^*$  ( $L_{19}$ ).

<span id="page-5-0"></span>

Figure 6. P-chiral bidentate phosphine ligands.

Phosphoramidite ligands were highly reactive and enantioselective in Rh-catalyzed asymmetric hydrogenation;<sup>96</sup> they also performed very well in Hayashi−Miyaura reaction because of their strong  $\pi$ -accepting properties.<sup>39,97-103</sup> For th[e R](#page-22-0)CAA of phenylboronic acid to 2-cyclohexenone, H<sub>8</sub>−BINOL-based ph[o](#page-21-0)sphoramidite (S)- $L_{22}$  proved to [be the](#page-22-0) most efficient.<sup>100</sup> Introducing the inexpensive methyl deoxycholic ester 23 as the source of chirality in phosphite  $L_{23}$  validated the sim[ilar](#page-22-0) efficiency compared with  $L_{24}$  (Figure 7).<sup>104,105</sup>



Figure 7. Monodentate phosphorus ligands.

The electron pair in N-heterocyclic carbenes (NHCs) makes them act as strong  $\sigma$ -donors. Despite their relatively weak π-acceptor property, NHC ligands still showed great catalytic activity in RCAA reaction. For example, the cyclophane-based NHC ligand  $L_{25}$  was a quite unique monodentate ligand and proved to be powerful for Hayashi-Miyaura reaction.<sup>106,107</sup> Remarkably, only 1 equiv of chiral NHC relative to rhodium was needed, as most monodentate ligands required 2 e[quiv to](#page-22-0) rhodium to achieve high selectivity (Figure 7).

2.4.2. Diene Ligands. As a result of two synergistic interactions of  $\sigma$ -donation from olefin to rhodium and  $\pi$ -back-donation from rhodium to olefin, structurally diverse chiral dienes should be excellent steering ligands. Indeed, the Rh(I)/chiral diene catalytic systems have proved to be one of the best catalytic methods for Rh-catalyzed asymmetric 1,4 addition because of the strong  $\pi$ -accepting ability of diene ligands. Since the first application of chiral diene  $L_{26b}$  in RCAA by Hayashi and co-workers,<sup>108</sup> a variety of bicyclic diene scaffolds have been successfully applied in this useful transformation (Figure 8).47,1[09](#page-22-0)<sup>−</sup><sup>117</sup> Almost at the same time, Carreira and co-workers reported that chiral diene is also effective for Ir-catalyz[ed](#page-6-0) [all](#page-21-0)[ylic su](#page-22-0)bstitution.<sup>118</sup> These independent discoveries have promoted intense research efforts in metal-catalyzed transformations, which were e[lega](#page-22-0)ntly reviewed by Carreira in  $2008.<sup>11</sup>$ 

As illustrated in Figure 8, all alkylated variations in chiral bicyclo[2.2.1]hepta[dien](#page-21-0)e scaffold, such as Me-  $(L_{26a})$ , Bn- $(L_{26b})$ , Cy-  $(L_{26d})$ , *i*-Bu-  $(L_{26e})$ , and allyl-  $(L_{26f})$  showed

excellent enantioselectivities (95−96% ee); and phenyl variation  $L_{26c}$  exhibited a higher enantiomeric excess (ee) value.<sup>119,120</sup> Chiral-bridged dienes  $L_{27}$ , which were easily prepared through catalytic enantioselective Diels−Alder reactio[n usin](#page-22-0)g the CBS catalyst, also displayed excellent selectivities (96–98% ee).<sup>121</sup> Other chiral bridged dienes, including bicyclo[2.2.2]octadiene variations  $(L_{28}, L_{29}, L_{30}, L_{31}, L_{32})$  $L_{32}$ ,  $L_{33}$ , and  $L_{34}$ ),<sup>110,11[8,12](#page-22-0)1–129</sup> bicyclo[3.3.1] nonadiene variations  $(L_{35}$  and  $L_{36})$ ,  $^{130,131}$  and bicyclo[3.3.2] decadiene variation  $L_{37}^{131}$  all [gave excellent](#page-22-0) enantiomeric excesses for Hayashi−Miyaura reaction[. Both](#page-22-0) optically pure enantiomers of bicyclo<sup>[3.3.0]</sup>octadiene-based chiral dienes  $L_{38}^{113}$  and  $L_{39}^{132}$ could be easily prepared through lipase-catalyzed kinetic resolution in >10 g scale. Hydrophilic diene [lig](#page-22-0)and  $L_{39}$  [is](#page-22-0) quite soluble in water (solubility: 5 mg/mL) and for the first time successfully promoted the RCAA in aqueous media within diene series.<sup>132</sup>

The Ph-dbcot  $L_{40}$ <sup>112</sup> and 1,5-Ph-cod  $L_{41}$ <sup>47</sup> are achiral, but the correspondi[ng c](#page-22-0)ationic rhodium complexes are chiral and both enantiomers can be [res](#page-22-0)olved. The opticall[y p](#page-21-0)ure Rh-complexes with  $L_{40}$  and  $L_{41}$  gave the moderate enantioselectivity in RCAA. Utilizing the simple and flexible acyclic chiral 1,5-dienes  $L_{42}$ <sup>133</sup>  $L_{43}$ ,<sup>133</sup>  $L_{44}$ ,<sup>133</sup> and  $L_{45}$ <sup>134</sup> as steering ligands for Hayashi– Miyaura reaction, good to excellent yields and ee were achie[ved](#page-22-0) (Fi[gure](#page-22-0) 8).

2.4.3. Bissulfoxide Ligands. Bis-sulfoxides are a newly rising family o[f](#page-6-0) ligands in homogeneous catalysis by coordinating Rh with the sulfur atoms.<sup>135,136</sup> Among them, chiral  $bis(p$ -tolylsulfoxide) ligands  $L_{46}^{137}$  and  $L_{47}^{138}$  were found to be outstanding ligands for H[ayashi](#page-22-0)−Miyaura reaction, demonstrating near-perfect enantiose[lect](#page-22-0)ivities ov[er](#page-22-0) various cyclic  $\alpha$ , $\beta$ -unsaturated ketones (Figure 9). A comparison of the X-ray crystal structures of  $\left[\text{RhCl}((R)-\text{BINAP})\right]_2$ ,  $\left[\text{RhCl}(\mathbf{L}_{28a})\right]_2^{139}$  and  $[RhCl(L_{46})]_2$  suggeste[d t](#page-6-0)hat the ligating nature of bis-sulfoxides might stay somewhere between that of diene [and](#page-22-0) bisarylphosphine ligands. Subsequently, a series of axially chiral bis(tert-butylsulfoxide) ligands  $L_{49a}$ - $L_{51a}$  and bis(p-tolyl-sulfoxide) ligands  $L_{49b}$ - $L_{51b}$  were synthesized, in which  $L_{49b}$ - $L_{51b}$ bearing p-tolylsulfinyl group proved to be remarkably efficient ligands for Hayashi−Miyaura reaction with perfect enantioselectivity (>99% ee).<sup>140</sup> A simple and readily prepared chiral bis(tert-butylsulfoxide) ligand L<sub>48</sub> also afforded the corresponding RCAA product[s in](#page-22-0) excellent yields and enantioselectivities.<sup>141</sup>

An emerging class of chiral heterodisulfoxide ligands  $L_{52}$ - $L_{56}$ cont[ain](#page-22-0)ing tert-butyl- and aryl-sulfoxides within a rigid benzene scaffold were developed and evaluated in the RCAA to chromenone  $(1h)$ ,<sup>142,143</sup> which is one of the most challenging subjects in this field. The simple bis(tert-butylsulfoxide) ligand L48 is still the mo[st activ](#page-22-0)e and selective ligand in this subseries (Scheme 14).<sup>141-143</sup> The highly electron-poor chiral diphosphine ligand  $L_{12c}$  was also applicable to this reaction, affording excellent [yiel](#page-6-0)[d and en](#page-22-0)antioselectivity.<sup>144</sup>

2.4.4. Phosphine-Olefin, -Nitrogen, -Oxygen, -Sulfur, or -NHC Hybrid Ligands. Hybrid liga[nds](#page-22-0) contain two different coordinating centers, with a phosphorus center to tune the steric and electronic properties and another functionality (such as an olefin, nitrogen, oxygen, sulfur, and NHC) to stabilize the Rh-complex. Phosphines coordinate more strongly to late transition metals than alkenes do; thus, the latter in phosphineolefin hybrid ligands are just to dilute the intense stickiness of the former to rhodium. Initially, the chiral phosphine-olefin ligand  $L_{57}$  was applied in the RCAA to enones,<sup>145</sup> and exhibited

<span id="page-6-0"></span>

Figure 8. Chiral diene ligands.



 $L_{49a}$ , trace  $L_{50a}$ , trace  $L_{51a}$ , trace (M, S, S)-p-Tol-MeO-BIPHESO (M, S, S)-p-Tol-MeO-SYNSO (M, S, S)-p-Tol-MeO-SEGSO  $L_{49b}$ , 98%, >99% ee  $L_{50b}$ , 89%, >99% ee  $L_{51b}$ , 87%, >99% ee For a:  $R = t$ -Bu; For b:  $R = 4$ -CH<sub>3</sub>-C<sub>6</sub>H<sub>4</sub>



higher catalytic activity than a  $\mathrm{Rh}/\mathrm{cod}$  catalyst.  $^{146}$  Subsequently, phosphine-olefin  $L_{58}^{147}$  and  $L_{59}^{148}$  amidophosphine-olefin  $L_{60}^{111,149,150}$  and carbohydrate phosphine[-ole](#page-22-0)fin  $L_{61}^{151,152}$ were developed and [sh](#page-22-0)owed goo[d a](#page-22-0)ctivities and ena[ntiose](#page-22-0)lec[tivities in](#page-22-0) Hayashi−Miyaura reaction (Figure 10).



Phosphine-nitrogen hybrid ligand  $L_{62}^{93}$  did not induce any enantioselectivity in RCAA, while the L-proline-derived



Figure 10. Phosphine-olefin, -nitrogen, -oxygen, -sulfur, or - NHC hybrid ligands.

phosphine-oxygen ligand  $L_{63}^{153-155}$  and phosphine-sulfoxide ligand  $L_{64}^{156,157}$  were found to be very effective. Phosphine-NHC hybrid ligand is a fairly [new fam](#page-22-0)ily and has been relatively little expl[ored in](#page-22-0) RCAA. Helmchen and co-workers showed that phosphine-NHC hybrid ligand  $L_{65}^{158}$  provided high yields and enantioselectivities in RCAA of arylboronic acids to enones and  $\alpha$ , $\beta$ -unsaturated esters (Figure 10)[.](#page-22-0)

2.4.5. Oxazoline-Olefin Hybrid Ligands. The oxazoline moiety is a ubiquitous, privileged structure in chiral ligands. With the idea that the combination of  $\eta^2$ -binding olefins with oxazolines would produce new coordination geometries and possibilities, Glorius and co-workers first introduced a novel class of oxazoline-olefin ligand  $L_{66}$  into RCAA reaction and demonstrated both modularity and versatility.<sup>159</sup> In contrast to many other classes of chiral ligands, this series are especially appealing because the steric and electronic [pro](#page-23-0)perties of the olefin components can be easily tuned. Subsequently, Franzen reported the synthesis of a novel oxazoline-indoleolefin ligand  $L_{67}$  and utilized it in Hayashi–Miyaura reaction (Figure 11).<sup>160</sup>



Figure 11. Oxazoline-olefin hybrid ligands.

2.4.6. Sulfoxide-Olefin Hybrid Ligands. Sulfoxide-olefin hybrids are also introduced as a new family of chiral heterobidentate ligands in RCAA. Knochel and co-workers first developed a concise synthesis of a couple of sulfoxideolefin hybrid ligands  $L_{68}$  and  $L_{69}$ , and applied them in Hayashi−Miyaura reaction to furnish the chiral products in excellent yields, with equally high enantioselectivities and opposite stereoconfigurations (Figure 12).<sup>161</sup> Interestingly, a new class of chiral sulfoxide-olefin ligands  $L_{70}$ <sup>162,163</sup> could induce reverse enantioselectivity by simply [ch](#page-23-0)anging the substitutions at the olefin moiety ( $L_{70b}$  [vs](#page-23-0)  $L_{70c}$  or  $L_{70d}$  vs  $L_{70e}$ ).





In other words, both enantiomers of the product can be synthesized by choosing different olefin-substituted  $(R)$ - $L_{70}$ . Such a great advantage can potentially eliminate the process to prepare the ligands in both enantiomeric pure forms. At the same time, Xu and co-workers also reported this class of sulfoxide-olefin ligands  $L_{71}$  for the same reaction.<sup>164</sup> Afterward, they demonstrated that the simple and readily available chiral sulfinamide-olefins  $L_{72}$ - $L_{74}$ <sup>165</sup> also displayed [grea](#page-23-0)t catalytic activities and enantioselectivities in Hayashi−Miyaura reaction. It is noteworthy that the su[lfur](#page-23-0)-stereogenic center  $(L_{74a}$  vs  $L_{74b})$ was the key chiral directing group and the two carbon chiralities seemed to be unnecessary in achieving excellent enantioselectivity. Facile preparation of these ligands is a remarkable advantage. The chiral sulfinamide-olefin ligands  $L_{72}$ - $L_{74}$  can be conveniently synthesized by using Zn-mediated allylation of chiral N-tert-butanesulfinylimines at room temperature, providing both enantiomers in high diastereo- and enantioselectivities.166−<sup>169</sup> Du and co-workers independently described the development of similar N-sulfinyl-based chiral sulfur-olefin

ligands  $\left( \mathbf{L}_{7\mathbf{5a}}\text{ and } \mathbf{L}_{7\mathbf{5b}}\right) ^{170}$  and their successful application in the same reaction. The carbon chiral center was also observed to have no impact on th[e en](#page-23-0)antioselectivity in RCAA. The ease of synthesis and needless consideration of the carbon chirality make this type of ligand attractive and promising for asymmetric catalysis. More recently, chiral sulfinimide-olefin ligand  $L_{75c}$ , with a single chiral sulfur atom, also exhibited promising activity and enantioselectivity in Hayashi−Miyaura reaction.<sup>1</sup>

# 3. RCA[A T](#page-23-0)O ALKENES

3.1. RCAA to  $\alpha$ , $\beta$ -Unsaturated Aldehydes. Rh-catalyzed conjugate addition to  $\alpha$ , $\beta$ -unsaturated aldehydes exhibited a special challenge because of the existence of a highly reactive aldehyde group, which can undergo 1,2-addition either in competition with 1,4-addition (path b vs path a) or further conversion of the RCAA product 25 to give alcohol 26 (Scheme 15). This competitive process is highly affected by the

Scheme 15. Competing Addition Reaction Pathways to Enals



properties of the ligands to afford different adducts.<sup>172</sup> In the addition of phenylboronic acid (2a) to cinnamaldehyde (24a), the  $t$ -Bu<sub>3</sub>P complex yielded the 1,2-addition pro[duc](#page-23-0)t  $27$ aa in 90% yield (Scheme 16), whereas the diene system afforded

#### Scheme 16. Ligand Controlled 1,4- or 1,2-Addition Selectivity



1,4-addition product 25aa in 88% yield without any 1,2-adduct even in the presence of excessive amount of phenylboronic acid (2 equiv). This interesting observation opens a door for the Rh(I)/chiral diene-catalyzed asymmetric conjugate addition to enals.

In 2005, Carreira and co-workers first presented a Rh(I)/ chiral diene  $(L_{31f})$ -catalyzed enantioselective 1,4-addition of arylboronic acids to enals.<sup>125</sup> The reaction was conducted in MeOH-water, giving the desired RCAA product 25ab in 80% yield and 92% ee. Conve[ntio](#page-22-0)nal ligands  $(R)$ -BINAP  $L_1$  and phosphoramidite  $(R)$ -L<sub>76</sub> gave poor results (only 33% and 19%) yield, respectively) although the former provided the desired 1,2-adduct with 89% ee (Scheme 17). However, in the presence





of chlorosilane,  $Rh(I)/(R)$ -BINAP proved to be a powerful catalytic system in the asymmetric 1,4-addition of arylzinc chlorides to enals, affording the corresponding 3,3-diarylpropanals with perfect enantioselectivities (98-99% ee).<sup>48</sup> Almost at the same time, the chiral dienes  $L_{26b}$  and  $L_{28b}$  were successfully applied in the same reaction by Hayashi and c[o](#page-21-0)workers.<sup>122</sup>

3.2. RCAA to  $\alpha$ , $\beta$ -Unsaturated Esters.  $\alpha$ , $\beta$ -Unsaturated esters ar[e ex](#page-22-0)cellent substrates for RCAA, providing only the 1,4 addition products. For linear substrates 28, the reactions proceeded smoothly when a more reactive arylboron reagent (LiArB(OMe)<sub>3</sub>, 10) was applied (Scheme 18).<sup>23,73</sup> Interestingly,



replacing water with 1 equiv of isopropanol could significantly reduce the protodeboration of arylboronic acids, and scale-up of 29ag (25 kg) was successfully obtained by using this improved condition.<sup>173</sup> Highly efficient RCAA to  $\beta$ , $\beta$ -disubstituted linear enoates 30, providing quaternary stereocenters at the  $\beta$ -position of esters, [was](#page-23-0) recently achieved by employing chiral diene  $L_{33b}$  as the ligand.<sup>174</sup>

RCAA to  $\beta$ -aryl substituted linear enoates 32 have been investigat[ed](#page-23-0) with Rh(I)/chiral diene  $(L_{31f})$  complex<sup>175</sup> or a cationic Rh(I)/chiraphos ( $L_{77}$ ) system<sup>72</sup> for the enantioselective preparation of  $β$ -diaryl esters 33 (Scheme 1[9\).](#page-23-0) Both catalytic systems proved to be quite [ve](#page-21-0)rsatile and functional group-tolerant. This method was successfully applied [as](#page-9-0) the key

<span id="page-9-0"></span>

enantioselective step in the asymmetric synthesis of two endothelin receptor antagonists 34 and 35.

 $Rh(I)/(S)$ -BINAP system was highly efficient in the RCAA to cyclic enoates  $36^{23,73}$  and  $Rh(I)/(R)$ -Segphos system presented excellent catalytic activity in the RCAA to coumarins  $38.176$  This conversion [was](#page-21-0) subsequently extended to the asymmetric synthesis of  $(R)$ -tolterodine  $(40)$ . Because of its strong  $\pi$ -[acce](#page-23-0)pting ability, the electron-deficient ligand  $L_{12c}^{86}$  exhibited a higher catalytic activity in the RCAA to the cyclic enoates although excessive arylboronic acids (10 equiv) w[ere](#page-22-0) required (Scheme 20).





The RCAA to enoates bearing  $\gamma$ - or  $\beta$ -N-phthaloylamino acrylates provided a useful approach to synthesize chiral  $\gamma$ - or  $\beta$ -amino acids. Both bisphosphorus ligand (R)-BINAP  $(L_1)^{177,178}$ and bis-sulfoxide ligand  $L_{48}^{156,157}$  were effective for the  $\gamma$ -Nphthaloylamino acrylate 41[,](#page-22-0) [with](#page-22-0) the latter showing [higher](#page-23-0) catalytic activity. These catalytic systems were applied to the asymmetric synthesis of  $(S)$ -Baclofen  $(43)$  and  $(S)$ -Rolipram (44). RCAA of arylboronic acids to  $\beta$ -N-phthaliminoacrylate ester 45 by using  $Rh(I)/chiral$  diene  $(L_{32b})$  complex toward the synthesis of β-amino acids was realized, giving β-aryl-β-Nphthaloylamino acid esters in high yields and enantioselectivities (Scheme 21). $179$ 





Diastereoselective RCAA was also explored with chiral enoate 47 as the substrate, providing a useful asymmetric synthesis of bicyclopyrrolizidinones 48.<sup>180</sup> Chiral diene ligand  $(S, S, S)$ -L<sub>31f</sub> enhanced the diastereoselectivity in the formation of  $(S,R)$ -48, while its enantiomeric ligand  $(R,R,R)$ - $L_{31f}$  reversed the diastereoselectivity of the RCAA reaction to give  $(S, S)$ -48. This result suggests the RCAA process should be under ligand control. This diastereoselective RCAA methodology was applied in a concise asymmetric synthesis of isaindigotidione 50 (Scheme 22).<sup>181</sup>

Di-tert-butyl fumarate (51) is less reactive compared with general α,β-[unsa](#page-10-0)t[ura](#page-23-0)ted esters in the RCAA with phenylboronic acid. Traditional bisphosphine ligand  $L_1$  and phosphoramidite ligand  $L_{76b}$  gave poor yield and enantioselectivity for this transformation, while the bulky chiral diene ligand  $L_{26g}$  could improve the process, affording the desired product 52 in 90% yield and 90% ee (Scheme  $23$ ).<sup>182</sup>

<span id="page-10-0"></span>



Scheme 23. RCAA of Phenylboronic Acid to Fumarate



3.3. RCAA to  $\alpha$ , $\beta$ -Unsaturated Amides. RCAA of linear  $\alpha$ , $\beta$ -unsaturated amide 53 with arylboronic acids performed similarly well despite its weaker reactivity than enones or enoates.<sup>64</sup> Replacing ArB(OH)<sub>2</sub> with ArBF<sub>3</sub>K (9) as the arylboronic source could significantly increase the overall reaction yield ([Sch](#page-21-0)eme 24).<sup>183</sup> Moderate yields and high enantioselectivities were obtained in RCAA to cyclic  $\alpha$ , $\beta$ -unsaturated- $\delta$ lactam 55.<sup>22</sup> Introd[ucin](#page-23-0)g N-Boc protecting group into the cyclic  $\alpha$ , $\beta$ -unsaturated amide 57 remarkably improved its reactivity,  $^{184}$ affording [β](#page-21-0)-substituted-γ-lactams with excellent yields and enantioselectivities. This highly efficient method was elega[ntly](#page-23-0) exemplified by the synthesis of two chiral drug molecules (R)- Baclofen  $(43)$  and  $(R)$ -Rolipram  $(44)$ .

Because of its low reactivity, the RCAA of N-benzyl maleimide (59) with phenylboronic acid gave only moderate enantioselectivity using bisphosphine ligand (R)-BINAP  $(L_1)$ .<sup>145</sup> Bulky chiral diene ligand  $L_{26g}$ <sup>145,182</sup> afforded  $\alpha$ phenylsuccinimide 60 in a higher yield, but still with low enan[tios](#page-22-0)electivity. Further exploration revea[led](#page-22-0) [tha](#page-23-0)t phosphorusolefin hybrid ligands  $L_{57}$ ,  $^{145,146}$  and  $L_{58}$ ,  $^{147}$  produced the desired adduct in excellent yields and enantioselectivities (Scheme 25). RCAA to [substi](#page-22-0)tuted ma[leim](#page-22-0)ides was also examined, leading to high regioselectivity and enantioselectivity. Hayashi and co-workers also reported a highly diastereoselective synthesis of axially chiral N-arylsuccinimides by using RCAA.<sup>182,185,186</sup>

Scheme 24. RCAA to Linear and Cyclic  $\alpha$ , $\beta$ -Unsaturated Amides



Scheme 25. RCAA of  $PhB(OH)$ <sub>2</sub> to N-Benzyl Maleimide



In summary, the rate of 1,4-addition of arylboron reagents to  $\alpha$ , $\beta$ -unsaturated carbonyl compounds in the RCAA process relies on the reactivity of the acceptor and its steric bulk, with the trend in the following order: enals > enones > enoates > enamides > fumarates > maleimides.<sup>8</sup> The steric bulk in proximity to the reactive unsaturated moiety will weaken the coordination of acceptor to rhodium a[nd](#page-21-0) accordingly decrease the reactivity in RCAA. In addition, the catalytic activity of Rh(I)/ligand systems will increase for the ligands with stronger  $\pi$ -accepting abilities.

3.4. RCAA to Nitroalkenes. Hayashi and co-workers initially demonstrated that cyclic nitroalkenes 61 were good substrates for the RCAA reaction, affording thermodynamically less stable cis-β-aryl substituted cyclic nitroalkanes 62 with excellent enantioselectivities (Scheme  $26$ ).<sup>187</sup> It is noteworthy

#### Scheme 26. RCAA to Cyclic Nitroalken[es](#page-23-0) [Rh(acac) $(C_2H_4)_2$ ]<br>(3 mol % Rh) NO, (S)-BINAP (3.3 mol%) +  $ArB(OH)<sub>2</sub>$ Dioxane, DMA or (5 or 10 equiv) DMF/H<sub>2</sub>O (10:1)  $n = 0, 1, 2$ 100 °C. 3 h 61  $\overline{2}$  $(1S.2S) - cis - 62$ 8 examples 73-93%, 73-99% ee  $cis/trans = 40/60 - 82/12$  $NAHCO<sub>3</sub>$ NO. EtOH, reflux  $(1S.2S)$ -cis-62ba  $(1R.2S)$ -trans-62ba  $cis/trans = 87/13$  $cis/trans = 3/97$ 98% ee 98% ee

that, as an example of 62ba, the stable *trans*-isomer could be easily obtained by treatment of the *cis-rich* mixture with sodium bicarbonate in refluxing ethanol.

This efficient methodology was then elegantly applied by a research group at Merck in the synthesis of CGRP receptor antagonist telcagepant 63 (Scheme 27).<sup>188</sup> The RCAA of





difluorophenylboronic acid (2j) to nitroalkene 61d provided the key chiral intermediate nitroalkane 62dj, which was further converted to telcagepant on a 2 kg scale. Bicarbonate was shown to be an effective activator in the RCAA reaction.

In 2010, Lin and co-workers reported the RCAA of aryl boronic acids to linear β-substituted 1-nitroalkenes and found the  $KHF_2$  additive was crucial to promote this catalytic reaction.<sup>189</sup> With  $L_{38}$  as the ligand, excellent enantioselectivities (95−97% ee) were obtained when sterically more hindered arylboro[nic a](#page-23-0)cids, such as 1-naphthylboronic acid and 2-tolylboronic acid, were used. To demonstrate the synthetic utility of this methodology, pharmaceutically interesting isoquinoline derivative 66 was easily prepared from the chiral  $\beta$ -arylnitroalkane product 65al (Scheme 28).

3.5. RCAA to  $\alpha,\beta$ -Unsaturated Phosphonates. The RCAA to alkenylphosphonates was first reported by Hayashi and co-workers.<sup>190</sup> Although alkenylphosphonates are usually less reactive than  $\alpha$ , $\beta$ -unsaturated carbonyl compounds, the RCAA reaction [pro](#page-23-0)ceeded smoothly by using  $Rh(I)/(S)$ -BINAP catalytic system with arylboroxines 6 as the boron source. Remarkably, the addition of 1 equiv of water (relative to boron





species) was required to achieve high yields. Consistent with the previous stereochemical model (cf. Scheme 7), the trans and cis geometries of alkenylphosphonate 67 afforded opposite enantiomers (S)-68 and (R)-68, respectivel[y](#page-2-0) (Scheme 29).

# Scheme 29. RCAA to  $\alpha$ , $\beta$ -Unsaturated Linear Phosphonates



3.6. RCAA to  $\alpha$ , $\beta$ -Unsaturated Sulfones. The initial RCAA to  $\alpha$ , $\beta$ -unsaturated phenylsulfones 69 was investigated by Hayashi and co-workers, who discovered that the sulfonyl group was eliminated through 1,2-Rh shift after the conjugate addition of aryltitaniums 70 as nucleophilic reagents, affording the desulfonylated alkenes 71 or 73 (Scheme 30).<sup>43</sup> However,

#### Scheme 30. RCAA to Alkenyl Phenylsulfones



replacing the aryltitanium reagents with arylboron reagents such as phenyl-9-BBN (12a) or triphenylcyclotriboroxane  $[(PhBO)<sub>3</sub>, 6a]$  failed to promote RCAA. Neither this type of cine substitution nor the 1,4-addition products were generated.

With the pyridyl-N to stabilize the Rh-complex G, the RCAA of arylboronic acids to  $\alpha$ , $\beta$ -unsaturated 2-pyridyl sulfones 74 was successfully achieved by using  $(S, S)$ -chiraphos  $(L_{77})$  as the chiral ligand, providing  $\beta$ -substituted sulfones in high yields  $\frac{1}{2}$  and enantioselectivities (Scheme 31).<sup>191−193</sup> The *trans*- and



 $cis$ -alkenylsulfones produced the opposite enantiomers  $(S)$ -75 and  $(R)$ -75 in RCAA reaction. The elimination of the sulfonyl group via Julia-Kociensky olefination<sup>194</sup> provides a novel approach to the enantioselective synthesis of allylic substituted trans-alkenes 76.

3.7. RCAA to Strained or Weakly-Activated Alkenes. Rhodium-catalyzed arylation of aryl boronic acids to a variety of strained oxabenzonorbornadienes has been demonstrated by Murakami and co-workers,<sup>195</sup> with the insertion of a strained alkene into a rhodium-aryl bond as a key step.<sup>196</sup> The best results were obtained wi[th](#page-23-0) a catalyst prepared by mixing  $[Rh(COD)Cl]_2$  with 2 equiv of  $P(OEt)_3$ . Lau[ten](#page-23-0)s and coworkers subsequently presented an asymmetric version of this RCAA reaction, which gave excellent enantio- and diastereoselectivities by using the  $[Rh(COD)Cl]_2/(R,S)$ -tBu-JOSI-PHOS- $L_{78}$  catalyst system (Scheme 32).<sup>197</sup> When mesoazabicycle 79 was used as the strained alkene, a chemodivergent desymmetrization occurred after the initi[al e](#page-23-0)nantioselective carbometalation step. The reaction brings an interesting approach to open diazabicyclo[2.2.1]heptanes enantioselectively to obtain arylated cyclopentenamines 80. An alternative reaction pathway was discovered for heteroaryl boronic acids, in which C−H insertion/1,4-Rh migration occurred to give hydroarylation products 81.<sup>198,199</sup> Recently, Lautens and coworkers demonstrated that Rh complex incorporating IBiox $[(-)$ -menthyl]  $(L_{79})$  [as](#page-23-0) [the](#page-23-0) ligand showed impressive selectivity, only allowing the hydroarylation of azabicycle regardless of aryl or heteroaryl boronic acids.<sup>200</sup>

A highly enantioselective RCAA of arylboronic acids to β-monosubstituted alkenylheteroarenes 83 [has](#page-23-0) been developed,<sup>201</sup> demonstrating the electron-deficient C=N-containing heteroarenes can activate the adjacent alkenes toward RCAA reacti[ons](#page-23-0) (Scheme 33). Very recently, Lam and co-workers also described another highly enantioselective RCAA of arylboronic acids to alkenyl-p-nitroarenes and alkenyl-p-cyano-m- (trifluoromethyl)arenes 85. <sup>202</sup> These reactions represented the first examples of catalytic asymmetric additions of airand moisture-stable orga[no-](#page-23-0)metallic reagents to alkenes activated by electron-deficient arenes.

Scheme 32. RCAA to Strained Alkenes







RCAA of arylboroxines 6 to cis-allylic diols 87 occurred using  $Rh(I)/chiral$  diene  $(L_{31f})$  catalyst system. This reaction proceeded through the syn-1,2-addition of arylrhodium species across the double bond and subsequent  $β$ -oxygen elimination to form 2-aryl-3-en-1-ols 88 with good enantioselectivities (Scheme 34). $^{203}$ 

Recently Hayashi and co-workers described the development of the R[CA](#page-13-0)[A o](#page-23-0)f arylboroxines 6 to borylalkenes  $89.^{204}$  The reaction afforded  $β$ -arylated alkylboron compounds 90 with high enantioselectivities by using Rh(I)/bisphosphine [lig](#page-23-0)ands such as  $(R)$ -Segphos $(L_{12e})$  and  $(R)$ -DTBM-Segphos $(L_{12i})$ . In this case, the 1,8-naphthalenediaminatoboryl group, B(dan), served as a key masking group for alkenylboronic acids during the RCAA, and could be further converted to hydroxyl group (Scheme 35).

<span id="page-13-0"></span>



Scheme 35. RCAA to Borylalkenes



3.8. 1,6-Conjugate Addition. Because of the multiple possible reaction pathways controlled by the substitution pattern of the  $\alpha$ , $\beta$ -γ, $\delta$ -diunsaturated carbonyl compounds, the selective 1,6-additions are especially challenging, as illustrated in Scheme 36.<sup>205</sup> For unhindered dienoates 92 ( $R^2 = H$  or Me),

Scheme 3[6. R](#page-23-0)h-Catalyzed 1,6-Addition or 1,4-Addition of Arylboronic Acids to 2,4-Dienoate Esters



the 1,6-addition product 93 was favored. However, when  $R^2$ was an aryl group, the 1,4-addition product 94 became predominant.

The regioselective 1,6-addition of  $\beta$ -substituted dienoates were first realized by using reactive arylzinc reagents 96 in combination with  $CISiMe<sub>3</sub>$ , providing  $(R)$ -97 in almost quantitative yields and good to excellent enantioselectivities (Scheme 37).<sup>50</sup> The enantioselective 1,6-addition of aryltitanate to alkynylenones 98 also proceeded by using  $Rh(I)/(R)$ -Segphos  $(L_{12e})$  $(L_{12e})$  catalytic system in the presence of ClSiMe<sub>3</sub>, affording enantiomerically enriched chiral allenes 100. <sup>46</sup> In 2010, Hayashi and co-workers developed a Rh-catalyzed asymmetric 1,6-addition of arylboronic acids to [lin](#page-21-0)ear enynamides 101 using a  $Rh(I)/chiral$  diene  $((S,S)-L_{32c})$ complex, producing axially chiral allenylsilanes 102 with high enantioselectivities.<sup>2</sup>

Scheme 37. Rh-Catalyzed Asymmetric 1,6-Addition onto α,β−γ,δ-Diunsaturated Carbonyl Compounds



# 4. RCAA TO ALDEHYDES AND KETONES

4.1. RCAA to Aldehydes. As shown in Scheme 38, the rhodium-catalyzed arylation of aldehyde and diarylzinc could

Scheme 38. Reactions of Aldehydes with Aryl-Rhodium Complexes to Form Diarylketones and Diarylmethanols



provide the desired product 111 under aqueous conditions. However, the ketone 108 was obtained under nonaqueous solvent system. Hartwig and co-workers looked into the mechanism of such a highly solvent-dependent process.<sup>207</sup> Despite the difficulty in isolating the triphenylphosphine-ligated  $aryl rhodium(I)$  complexes in a pure form,<sup>208</sup> a[ryl-](#page-23-0)rhodium compounds with a mixed phosphine and CO ligation sphere are usually stable. They did indeed obtain the [pure](#page-23-0) aryl-rhodium complex 105 by treatment of  $(PPh_3)_2Rh(CO)Cl$  with 1 equiv of diarylzinc in tetrahydrofuran (THF) at room temperature. The reaction of this stable complex 105 with aldehyde 106 led to the insertion of a carbonyl group into the aryl-rhodium

bond, generating rhodium-alkoxide 107. Under nonaqueous  $(C_6D_6)$  condition, the reaction went through  $\beta$ -elimination to produce the diarylketone 108 and release a new rhodiumhydride complex 109. This [Rh−H] complex was not very stable and subsequently decomposed under the reaction conditions to liberate hydrogen and form  $[Rh(\mu\text{-CO})(\text{PPh}_3)_2]_2$ (110). Because of the high stability of aryl-rhodium complex 105, the reaction could be conducted in aqueous media. By using a mixture of THF and water as solvent, the reaction exclusively generated the desired product, diarylmethanol 111 through hydrolysis (Scheme 38).

Phosphoramidite  $L_{80}$  is an effective chiral ligand in the RCAA of arylboronic acids to al[deh](#page-13-0)ydes, providing chiral diarylmethanols in high yields and up to 75% ee (Scheme 39).<sup>209</sup>



Chiral spiro-phosphite ligand  $L_{81}$  also proved to provide excellent yields and good enantiomeric excesses in RCAA of arylboronic acids to aldehydes.<sup>210</sup> Diene ligands were also applied in the RCAA reaction. Although the  $[RhCl((S,S)-])$  $\mathbf{L}_{26e})$ <sub>2</sub>-catalyzed 1,2-addition o[f p](#page-23-0)henylboronic acid to aryl aldehydes provided the desired alcohols in 98% yield, the enantioselectivity was quite poor (only 41% ee).<sup>120'</sup> In 2009, Hayashi and co-workers developed a novel  $C_2$ -symmetric tetrafluorobenzobarrelene ligand and applied it i[n th](#page-22-0)e RCAA of arylboronic acids to aryl aldehydes, affording chiral diarylmethanols in high yields and high enantioselectivities.<sup>211</sup> More recently, both chiral monophosphorus ligand  $(R)$ - $L_{82}^{212}$ bearing a di(trifluoromethyl)alcohol moiety and chiral N[HC](#page-23-0)ligand  $(S_p)$ - $L_{25}$ <sup>213</sup> demonstrated excellent catalytic activity [for](#page-23-0) asymmetric 1,2-addition of arylboronic acids to aldehydes. Remarkably, t[he](#page-23-0) fluoroalcohol moiety in  $(R)$ -L<sub>82</sub> played a pivotal role for the high enantioselectivity in Rh(I)-catalyzed transformation. However, only moderate enantioselectivity was observed in RCAA when using  $(S_p)$ - $L_{25}$  (with 0.03–0.3 mol % loading) as the ligand.

The chiral alcohol generated in RCAA can further react in situ with other functional groups in the molecule. As shown in Scheme 40, Hu and co-workers applied methyl 2-formylbenzoate (113) in the RCAA with various arylboronic acids; a series of chiral 3-substituted phthalides 114 (with up to 83% ee) were conveniently synthesized by using Rh(I)/SPINOLbased phosphite  $(L_{81})$  catalytic system.<sup>214</sup>

Scheme 40. RCAA to Methyl 2-Formylbenzoate (113)



4.2. RCAA to Ketones. Similar to aryl aldehydes, the carbonyl functionality activated by adjacent electron-withdrawing groups, like ester, amide, and trifluoromethyl, is also an excellent acceptor in RCAA. As shown in Scheme 41, the

#### Scheme 41. RCAA to Isatins



carbonyl group in isatins 115 could react with arylboronic acids through RCAA to produce biologically relevant 3-aryl-3 hydroxy-2-oxindoles 116 in high yields and enantioselectivities by using (R)-MeO-MOP  $(L_{83})$  as the ligand.<sup>215</sup>

Feringa and co-workers reported the catalytic asymmetric 1,2-addition of a series of arylboronic [ac](#page-23-0)ids to 2,2,2 trifluoroacetophenones 117. The reaction afforded the tertiary alcohols 118 in high yields (up to 96%) and good enantioselectivities (up to 83% ee) by using a  $Rh(I)/$ phosphoramidite catalyst (Scheme 42).<sup>216</sup>





Zhou and co-workers applied the chiral spirophosphite ligand (S)-L<sub>81b</sub> in asymmetric addition of arylboronic acids to  $\alpha$ ketoesters 119. Interestingly, the RCAA of 121 selectively took place on the activated carbonyl group rather than the alkenyl via a 1,4-addition process. This protocol provided a new enantioselective approach to the synthesis of 2-hydroxydiarylacetates 120 and alkenylarylacetates 122, which could be further derived to related  $\alpha$ -hydroxy carboxylic acids and vicinal diols bearing a tertiary chiral center (Scheme 43).<sup>217</sup>

#### 5. RCAA TO IMINES

Chiral diarylmethylamines and diarylmethanols are important structural motifs that are encountered in many pharmaceuticals and natural products.<sup>218</sup> Similar to the above RCAA of aldehydes/ketones to afford the diarylmethanols, imines are

#### <span id="page-15-0"></span>Scheme 43. RCAA to  $\alpha$ -Ketoesters



also suitable substrates for RCAA, representing one of the most straightforward ways to access chiral diarylmethylamines in high yields and stereoselectivities.

5.1. RCAA to Chiral Sulfinyl Imines. In 2005, Ellman and co-workers first reported the addition of arylboronic acids to both aromatic and aliphatic N-tert-butanesulfinyl imines 123, providing the chiral sulfinamides 124 with high diastereoselectivities. More importantly, the N-tert-butanesulfinyl group can be conveniently cleaved under mild acidic conditions that tolerate much sensitive functionality.<sup>219</sup> As shown in Scheme 44, the RCAA of aryl trifluoroborates 9 or aryl MIDA

Scheme 44. Diastereoselective RCAA to N-tert-Butanesulfinyl Imines



boronates 8 to N-tert-butanesulfinyl imines 123 also proceeded in excellent yields (up to 99%) and with very high diastereoselectivities  $(98.2 \text{ to } >99.1)$ .<sup>220</sup> Almost at the same time, Batey and co-workers independently reported the RCAA of arylboronic acids to chiral sulfinyli[min](#page-23-0)es in the absence of external chiral phosphine ligands. This substrate-controlled asymmetric process also afforded the desired products with excellent diastereoselectivities.<sup>221</sup>

N-Sulfinyl imino esters 125 stand out as stable, isolable compounds that can even be [ch](#page-23-0)romatographed on  $SiO<sub>2</sub>$ , and thus can serve as excellent starting materials for arylglycine synthesis.222−<sup>224</sup> In 2006, Ellman and co-workers applied them in RCAA of arylboronic acids to give a variety of N-tertbutanesu[lfinami](#page-23-0)do arylglycine esters 126 with excellent diastereoselectivities.<sup>225</sup> In 2009, Truong and co-workers developed an efficient diastereoselective RCAA of arylboronic acids to N-tert-but[ane](#page-23-0)sulfinyl trifluoromethyl imine 127, generating the corresponding sulfinamides 128 in good yields and excellent diastereoselectivities (up to 98% de).<sup>226</sup> This protocol provides a convenient method to prepare a variety of chiral trifluoroethylamine analogues 129 (Scheme 45[\).](#page-23-0)

# Scheme 45. Diastereoselective RCAA to N-tert-Butanesulfinyl Imino Esters



5.2. RCAA to Sulfonyl Imines. In 2004, Tomioka and coworkers reported the first RCAA of arylboronic acids and arylboroxines to N-tosylarylimines 130 using a chiral amidomonophosphane ligand (Scheme  $46$ ).<sup>227</sup> Almost at the same time,

#### Scheme 46. RCAA to N-Tosyl [Aldi](#page-23-0)mines



Hayashi and co-workers utilized bicyclo[2.2.2]octadiene ligands in the same transformation.<sup>119,131,139</sup> In 2006, Zhou and coworkers applied the rhodium complex of monodentate spirophosphite (S)-ShiP ( $L_{81c}$ ) in [RCAA. T](#page-22-0)he reaction proceeded in aqueous toluene to give diarylmethylamines 131 in good yields and enantioselectivities (up to  $96\%$  ee).<sup>228</sup>

Bicyclo[3.3.0]diene  $(L_{38})$  was found to be an excellent ligand for the arylation of sulfonyl i[mine](#page-23-0)s. A wide variety of N-tosylarylimines with diverse steric and electronic properties successfully reacted with several arylboronic acids, providing the corresponding N-diarylmethyltosylamides with a very

narrow range of excellent enantioselectivities (98−99% ee). The electronic and steric nature of phenyl ring at either imines or arylboronic acids apparently had no influence on the enantioselectivity. Furthermore, tosylimines derived from heteroaryl aldehydes such as furanyl, thiophenyl, and indolyl aldehydes were also suitable substrates for this RCAA reaction.<sup>229,230</sup> In 2010, a new class of monosubstituted  $C_1$ symmetric diene ligands with a DCP (dicyclopentadiene) backbon[e was](#page-23-0) developed by Lin and co-workers. The reaction proceeded smoothly to give the corresponding N-diarylmethyl tosylamides with excellent yields (98−99%) and high enantioselectivities (90-96% ee).<sup>231</sup> More recently, Hayashi designed and synthesized a novel chiral phosphine-olefin hybrid ligand  $L_{85}$ , an interesting bidentat[e bi](#page-23-0)nding to rhodium, which also resulted in high enantioselectivities in RCAA to imines 130 (Scheme  $46$ ).<sup>232</sup>

In spite of the considerable advancements, the imine substrate[s in](#page-15-0) [RC](#page-23-0)AA reaction are usually limited to aromatic imines. The major challenge in this case is that aliphatic imines tend to undergo imine-enamine tautomerization, decomposition, and self-condensation under common RCAA reaction conditions. In 2011, Lin and co-workers successfully extended the RCAA reaction to aliphatic N-tosylimines 132 using chiral Rh(I)/diene complexes as catalyst under neutral reaction conditions.<sup>233</sup> The bicyclo<sup>[3.3.0]</sup>octadiene  $L_{38}$  proved to be the superior ligand for this transformation, providing the desired products 1[33](#page-23-0) in exceptionally high enantioselectivity (typically ≥99% ee). With chloro-substituted imines 134 as substrates in RCAA, chiral 2-aryl pyrrolidines and piperidines were achieved in a one-pot procedure (Scheme 47). One drawback in this





reaction is the difficulty in removing the tosyl group from nitrogen of the products 133 and 135. It has been reported that the use of 4-nitrobenzenesulfonyl (Ns) group<sup>234-236</sup> in place of tosyl group will facilitate the deprotection reaction.

Several groups explored the RCAA of N-[4-ni](#page-23-0)t[rob](#page-23-0)enzenesulfonylimines. Unsurprisingly, high enantioselectivities and high catalytic activities were observed by using chiral diene ligands  $(L_{35}, ^{130} L_{33b}, ^{129}$  and  $L_{38}^{233})$ . The 4-nitrobenzene-sulfonyl (Ns) group could be readily removed from the products 137 without race[miz](#page-22-0)ation [or](#page-22-0) any sid[e re](#page-23-0)actions (Scheme 48).

Feringa and co-workers introduced a small and cheap N,Ndimethylsulfamoyl protecting group into the catalytic asymmetric synthesis of diarylmethylamines by RCAA reaction. The addition of 138 with aryl boronic acids afforded high

Scheme 48. RCAA to N-(4-Nitrobenzene)sulfonyl Aldimines



enantioselectivities (up to 95% ee) and high yields (up to 98%) by using a  $Rh(I)/phosphoramidite$  system.<sup>237</sup> The protecting group could be easily removed by microwaveassisted transamination, representing a versatile and [sel](#page-23-0)ective transformation to prepare chiral diarylmethylamines. Later on, Du and co-workers utilized bisnaphthyl-based chiral diene as the steering ligand for the enantioselective arylation of N,Ndimethylsulfamoyl-protected aldimines with arylboronic acids, providing the desired products 139 in moderate to good yields and up to 84% ee (Scheme 49). $238$ 





Bis-sulfamyl imines 140 are potentially ideal substrates for rhodium-catalyzed asymmetric additions of arylboron nucleophiles as they provided (i) near perfect enantioselectivities (11 examples, 98−99+% ee), (ii) good to excellent diastereoselectivities (10−32:1 rac/meso), and (iii) high functional group tolerance in the removal of protecting group via mild heating in aqueous pyridine (Scheme  $50$ ).<sup>239</sup>





Recently, Hayashi and co-workers successfully extended the RCAA reaction to less reactive N-tosyl ketimines 142 using sodium tetraarylborates or aryl trifluoroborates as boron reagents. By employing chiral diene  $(R,R)$ - $L_{28c}$  as the ligand,

the reaction produced various chiral amine derivatives 143 possessing a tetrasubstituted carbon stereocenter in high yields and enantioselectivities (Scheme  $51$ ).<sup>240,241</sup>

#### Scheme 51. RCAA to N-Tosyl Keti[mines](#page-24-0)



5.3. RCAA to Phosphonyl Imines. In 2005, Ellman and co-workers reported the highly enantioselective addition of arylboronic acids to N-diphenylphosphinoyl benzaldimines 144 using chiral ligand  $L_{88}$ .<sup>219</sup> Later on, arylboroxines were also successfully applied in this RCAA reaction by sterically tuning the diphenylphosphoru[s m](#page-23-0)oiety to a  $di(o$ -tolyl)phosphorus based ligand  $L_{84b}$ .  $^{242,243}$  Both chiral ligands provided high enantioselectivities in the preparation of diaryl derivatives 145. It is noteworthy [that](#page-24-0) [N](#page-24-0)-diphenylphosphinoyl group can be readily cleaved under mild acidic conditions that tolerate much sensitive functionality (Scheme 52).







Azomethine imines 146 are also suitable substrates for RCAA reaction. Hayashi and co-workers successfully developed a RCAA system by using sodium tetraarylborates as the nucleophile. With chiral diene  $(R, R, R)$ - $L_{33b}$  as the ligand, the reaction generated a series of chiral 1-(diarylmethyl) pyrazolidin-3-ones 147 in excellent yields and high enantioselectivities (Scheme  $53$ ).<sup>244</sup>

Very recently, Nadeau and co-workers developed the first RCAA of arylboronic [acid](#page-24-0)s to N-benzylnicotinate salts 148.





A variety of 6-substituted dihydropyridines 149 were isolated in good yields and excellent enantioselectivities (Scheme 54).<sup>245</sup>



### 6. RCAA-TANDEM REACTION

In the RCAA of organoboron reagents to electron-deficient alkenes described above, the insertion of unsaturated bonds to [Rh−Ar] intermediate always led to a chiral rhodium species, which could be further trapped by electrophiles other than a proton. In the case of using  $\alpha$ , $\beta$ -unsaturated ketone as the acceptor for RCAA, a chiral Rh-enolate is formed. This sequential "RCAA/enolate trapping" procedure represents one of the most effective strategies in asymmetric catalytic transformations, allowing for the formation of multiple C−C bonds in an atom-economic manner.<sup>246</sup> On the other hand, arylrhodium(I) species can preferentially undergo facile 1,2-syn addition across C−C triple bond in [the](#page-24-0) presence of alkene function within the molecule, the resulting alkenylrhodium(I) intermediate provides another entry for a tandem reaction to the neighboring C−C double bond. These tandem processes have been extensively discussed in the literature.<sup>247–249</sup>

6.1. Tandem RCAA/Aldol Reaction. In 2002, Hayashi and co-workers developed an elegant three-com[ponent t](#page-24-0)andem RCAA/aldol reaction, which was realized with high syn selectivity (syn/anti up to 21.4:1) by using  $ArB(9-BBN)$  as arylboron reagents and  $[Rh(OMe)(COD)]_2$  as a catalyst.<sup>250</sup> Applying  $[Rh(OH)((S) - BINAP)]_2$  as the chiral catalyst, the asymmetric version of the above reaction from tert-butyl vi[nyl](#page-24-0) ketone (150a),  $(4-F-C<sub>6</sub>H<sub>4</sub>)-B(9-BBN)$  (12b), and propanal (151b) gave optically active syn- and anti products. Despite the poor syn/anti selectivity (only 0.8:1), the syn-(4S,5R)-152abb and  $anti-(4R,5R)$ -152abb showed 41% ee and 94% ee, respectively (Scheme 55).

In the above case of using  $[Rh(OMe)(COD)]_2$  as the catalyst, the high syn [se](#page-18-0)lectivity suggested the cis-Rh enolate (cis-C) should have served as the predominant intermediate in the sequential aldol reaction. When  $[Rh(OH)((S)$ -BINAP)]<sub>2</sub> was used as the catalyst, both cis- and trans-Rh enolates were formed because of the highly skewed structure and huge hindrance from  $(S)$ -BINAP ligand,<sup>17</sup> generating a mixture of syn- and anti-aldol products (Scheme 56).

Hayashi and co-workers applied [a](#page-21-0) new combination system of ArB(9-BBN) (12) and  $[Rh(OMe)((S)$ -BINAP)]<sub>2</sub> in tandem RCAA/aldol reaction. Cyclic  $\alpha$ , $\beta$ -un[satu](#page-18-0)rated ketones could

#### <span id="page-18-0"></span>Scheme 55. Three-Component Tandem RCAA/Aldol Reaction



Scheme 56. Tandem RCAA/Aldol Reaction by  $\left[\text{Rh}(\text{OMe})(\text{COD})\right]_2$  or  $\left[\text{Rh}(\text{OH})((S)\text{-BINAP})\right]_2$ 



generate chiral boron enolates (S)-154 in  $\geq$ 96% ee, which successfully underwent further aldol reaction to give 155ba and 155ea with perfect regio- and diastereoselectivities (Scheme  $57$ ).<sup>39</sup>





In 2003, Krische and co-workers reported an intramolecular tandem RCAA/aldol reaction with high diastereo- and enantioselectivities.<sup>251</sup> The RCAA of phenylboronic acid 2a to a monoketoenone 156 generated an (oxa-π-allyl)rhodium species J-1, which [sub](#page-24-0)sequently attacked the ketone acceptor and further liberated the cyclic aldol products 157 under protic condition. Because of the high electrophilicity of the neighboring ketone, the subsequent aldol reaction was more kinetically favored than the hydrolysis of the  $(oxa-\pi-allyl)$ rhodium species J-1 (Scheme 58).

Scheme 58. Intramolecular Tandem RCAA/Aldol Reaction of Monoketoenones



In 2004, Krische and co-workers extended the above intramolecular tandem RCAA/aldol reaction to the desymmetrization and parallel kinetic resolution of diketoenones 158. This elegant tandem process provided interesting bicycle products 159, which embody four contiguous stereocenters, including two adjacent quaternary centers, with quantitative diastereoselection and high levels of enantiomeric excess (Scheme 59). $252$  This methodology enabled the rapid assembly





of complex polycyclic ring systems from simple precursors. For example, the RCAA/aldol tandem reaction of 158a allowed for a concise entry to optically pure seco-B ring steroid 159ab possessing a 14-hydroxy cis-fused C−D ring junction.

6.2. Tandem RCAA/Alkylation Reaction. The reaction of chiral boron- or rhodium-enolate with alkyl halides could give optically pure alkylated products. As shown in Scheme 60, the treatment of boron-enolate  $(S)$ -154 with *n*-butyllithium at −78 °C, followed by reaction with allyl bromide, gave ch[iral](#page-19-0) 2-allylcyclohexanone  $162$  as a single diastereoisomer.<sup>39</sup> Similarly, treatment of chiral titanium-enolate or zinc-enolate in the same manner could also lead to the corresponding chiral R[CA](#page-21-0)A/alkylation products.<sup>44,49</sup>

<span id="page-19-0"></span>Scheme 60. Tandem RCAA/Alkylation Reaction



6.3. Tandem RCAA/1, 2-Addition Reaction. Cyano group can also serve as an electrophilic acceptor to trap the rhodium-enolate. In 2007, Murakami and co-workers developed a new tandem reaction triggered by the RCAA of ArB(9- BBN) (12) to cyano-substituted  $\alpha$ , $\beta$ -unsaturated esters 163, providing chiral five-membered  $\alpha$ -enamino esters 164 with up to 95% ee (Scheme 61).<sup>253</sup> Initially, RCAA of arylrhodium(I)

Scheme 61. Tandem R[CAA](#page-24-0)/1,2 Addition onto a Cyano Group



species to 163 afforded the (oxa- $\pi$ -allyl)rhodium(I) intermediate K-1, which underwent intramolecular 1,2-addition onto the cyano moiety to form the N-rhodium $(I)$  imine K-2. Subsequent transmetalation of K-2 with ArB(9-BBN) (12) produced the N-boryl imine species K-3, which led to the final product 164 through hydrolysis/tautomerization. A related example with strained bicyclic alkene 165 and bifunctional (2 cyanophenyl)boronic acid 2k did demonstrate a  $\begin{bmatrix} 3 & + & 2 \end{bmatrix}$ annulation reaction, through the tandem RCAA/1,2-addition process, to afford an optically active exo-adduct 166 by using a rhodium/ $L_{31a}$  catalyst.<sup>254</sup>

Aryl-rhodium species can undergo syn-addition to dialkylacetylenes and subseq[uen](#page-24-0)tly cyclize through the intramolecular 1,2-addition to aldehydes or ketones. In 2005, Hayashi and coworkers described this tandem Rh-catalyzed arylation (RCA)/ 1,2-addition of arylboronic acids to alkynals and alkynones 167. The reaction successfully afforded chiral cyclic allylic alcohols 168 in high yields and enantioselectivities by applying a  $Rh(I)/$  $(S, S)$ -L<sub>28b</sub> catalyst (Scheme 62).<sup>255</sup> A related reaction between bifunctional (2-formylphenyl)boronic acid 2m and alkyne 169

# Scheme 62. Tandem RCA/1,2-Addition to an Aldehyde Group



generated the optically active 2-indenol 170 through the tandem  $RCA/1,2$ -addition process.<sup>256</sup>

6.4. Tandem Carborhodation/Conjugate Addition. Similar to the above process, aryl-r[hod](#page-24-0)ium species can undergo syn addition to a carbon−carbon triple bond, followed by cyclization through the intramolecular 1,4-conjugate addition to electron-deficient olefins. In 2005, Hayashi and co-workers developed a Rh-catalyzed arylative cyclization of alkynetethered  $\alpha$ , $\beta$ -unsaturated esters 171 with excellent chemoand enantioselectivities by the use of a chiral diene ligand  $(S, S)$ - $L_{28b}$  (Scheme 63).<sup>257</sup> Interestingly, Rh(I)/bisphosphine system

Scheme 63. Rh-[Cata](#page-24-0)lyzed Arylative Cyclization of Alkyne-Tethered Electron-Deficient Olefins



catalyzed the 1,4-addition of  $\alpha$ , $\beta$ -enoates more effectively than the arylation of alkynes, whereas a Rh(I)/diene catalyst exhibited high preference to the arylation of alkynes. This remarkably different behavior was probably due to the more electrophilic nature of a  $Rh(I)/d$ iene complex than a  $Rh(I)/d$ phosphine center.

Recently, this strategy was extended to alkyne-tethered 2-cycloalken-1-ones 173 and sodium tetraarylborates 13,

<span id="page-20-0"></span>affording spirocarbocycles 174 with quaternary spirocarbon stereocenters in excellent enantiomeric purity.<sup>258</sup> The arylrhodium species went through 1,2-syn insertion of the alkyne group in 173, giving an alkenylrhodium interm[edia](#page-24-0)te N-1. An interesting 1,4-Rh migration occurred to generate a new arylrhodium species N-2, which further reacted with the intramolecular enone group to afford oxa-π-allylrhodium intermediate N-3. Protonation of N-3 in an aqueous media provided the chiral spiro-cyclic product 174. Overall, the aryl groups in tetraarylborates functioned as surrogates of 1,2 dimetalloarenes and sequentially formed two new carbon− carbon bonds (Scheme 64).

# Scheme 64. Rh-Catalyzed Arylative Cyclization of Alkyne-Tethered 2-Cycloalken-1-ones



6.5. 1,6-Enyne Sequential Cyclization. Murakami and co-workers reported the  $Rh(I)/(R)$ -BINAP catalyzed cascade reaction of 1,6-enyne 175 with phenylboronic acid 2a to produce optically active cyclopentane 176 with excellent enantioselectivity (Scheme  $65)$ <sup>259</sup> The whole sequence

# Scheme 65. Rh-Catalyzed Arylati[ve](#page-24-0) Cyclization of 1,6- Enyne-Tethered Allylic Ether



involved 1,2-syn addition to the triple bond, 5-exotrig cyclization into the olefin, and  $\beta$ -oxygen elimination to give 176. The methoxyl group played a dominating role in the catalytic cycle, terminating the alkyl-rhodium species O-2 and regenerating the catalytically active methoxorhodium(I) species.

In the absence of a proper allylic leaving group as terminating function adjacent to the alkene moiety in 1,6-enyne, the resulting alkyl-rhodium species P-2 could not undergo either  $\beta$ oxygen elimination or  $β$ -hydrogen elimination because of the existence of a substitution group  $\mathbb{R}^2$ . Instead, the intermediate P-2 further reacted with an electrophilic ester group in the molecule, through a 5-exotrig cyclization, affording the optically active 2-norbornanone skeleton 178 with high enantioselectivities (Scheme 66).<sup>260</sup>

#### Scheme 66. Rh-C[atal](#page-24-0)yzed Arylative Cyclization of 1,6-Enyne



# 7. OUTLOOK

Since the initial breakthrough in the field of RCAA in 1998, distinct families of chiral ligands and various highly efficient asymmetric catalytic systems have been developed and established for different reaction substrates. With a large number of chiral ligands and optimized conditions available in achieving high stereoselectivity, the RCAA reaction is becoming more and more important in organic synthesis. However, the generation of aryl-rhodium species from direct C−H activation instead of through transmetalation and further extension of RCAA reaction to the nonactivated alkenes are still quite challenging in the field. Additional approaches in this aspect are expected with much anticipation indeed.

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#### ■ REFERENCES

(1) Hayashi, T. Synlett 2001, No. Special Issue, 879.

- (2) Hayashi, T. Russ. Chem. Bull., Int. Ed. 2003, 52, 2595.
- (3) Hayashi, T.; Yamasaki, K. Chem. Rev. 2003, 103, 2829.

(4) Yoshida, K.; Hayashi, T. In Modern Rhodium-Catalyzed Organic Reactions; Evans, P. A., Ed.; Wiley-VCH: Weinheim, Germany, 2005; Chapter 3, p 55.

<span id="page-21-0"></span>(5) Yoshida, K.; Hayashi, T. In Boronic Acids: Preparation and Applications in Organic Synthesis and Medicine; Hall, D. G., Ed.; Wiley-VCH: Weinheim, Germany, 2005; Chapter 4, p 171.

(6) Shintani, R.; Hayashi, T. In New Frontiers in Asymmetric Catalysis; Mikami, K., Lautens, M., Eds.; Wiley-Interscience: Hoboken N. J., 2007; Chapter 3, p 59.

(7) Shintani, R.; Hayashi, T. Aldrichimica Acta 2009, 42, 31.

(8) Berthon, G.; Hayashi, T. In Catalytic Asymmetric Conjugate Reactions; Córdova, A., Ed.; Wiley-VCH: Weinheim, Germany, 2010; Chapter 1, p 1.

(9) Edwards, H. J.; Hargrave, J. D.; Penrose, S. D.; Frost, C. G. Chem. Soc. Rev. 2010, 39, 2093.

(10) Johnson, J. B.; Rovis, T. Angew. Chem., Int. Ed. 2008, 47, 840.

(11) Defieber, C.; Grützmacher, H.; Carreira, E. M. Angew. Chem., Int. Ed. 2008, 47, 4482.

(12) Sakai, M.; Hayashi, H.; Miyaura, N. Organometallics 1997, 16, 4229.

(13) Takaya, Y.; Ogasawara, M.; Hayashi, T.; Sakai, M.; Miyaura, N. J. Am. Chem. Soc. 1998, 120, 5579.

(14) Hayashi, T.; Takahashi, M.; Takaya, Y.; Ogasawara, M. J. Am. Chem. Soc. 2002, 124, 5052.

(15) Kina, A.; Iwamura, H.; Hayashi, T. J. Am. Chem. Soc. 2006, 128, 3904.

(16) Kina, A.; Yasuhara, Y.; Nishimura, T.; Iwamura, H.; Hayashi, T. Chem.-Asian J. 2006, 1, 707.

(17) Ozawa, F.; Kubo, A.; Matsumoto, Y.; Hayashi, T.; Nishioka, E.; Yanagi, K.; Moriguchi, K. Organometallics 1993, 12, 4188.

(18) Miyaura, N. Bull. Chem. Soc. Jpn. 2008, 81, 1535.

(19) Matos, K.; Soderquist, J. A. J. Org. Chem. 1998, 63, 461.

(20) Itooka, R.; Iguchi, Y.; Miyaura, N. J. Org. Chem. 2003, 68, 6000.

(21) Chen, F. -X.; Kina, A.; Hayashi, T. Org. Lett. 2006, 8, 341.

(22) Senda, T.; Ogasawara, M.; Hayashi, T. J. Org. Chem. 2001, 66, 6852.

(23) Takaya, Y.; Senda, T.; Kurushima, H.; Ogasawara, M.; Hayashi, T. Tetrahedron: Asymmetry 1999, 10, 4047.

(24) Brak, K.; Ellman, J. A. J. Org. Chem. 2010, 75, 3147.

(25) Pucheault, M.; Darses, S.; Genêt, J. -P. Eur. J. Org. Chem. 2002, 3552.

(26) Pucheault, M.; Darses, S.; Genêt, J. -P. Tetrahedron Lett. 2002, 43, 6155.

(27) Darses, S.; Genêt, J. -P. Eur. J. Org. Chem. 2003, 4313.

(28) Molander, G. A.; Figueroa, R. Aldrichimica Acta 2005, 38, 49.

(29) Navarre, L.; Martinez, R.; Genêt, J. -P.; Darses, S. J. Am. Chem. Soc. 2008, 130, 6159.

(30) Batey, R. A.; Quach, T. D. Tetrahedron Lett. 2001, 42, 9099.

(31) Molander, G. A.; Biolatto, B. Org. Lett. 2002, 4, 1867.

(32) Molander, G. A.; Biolatto, B. J. Org. Chem. 2003, 68, 4302.

(33) Yuen, A. K. L.; Hutton, C. A. Tetrahedron Lett. 2005, 46, 7899.

(34) Gendrineau, T.; Genêt, J. -P.; Darses, S. Org. Lett. 2009, 11, 3486.

(35) Takaya, Y.; Ogasawara, M.; Hayashi, T. Tetrahedron Lett. 1999, 40, 6957.

(36) Hayashi, T.; Takahashi, M.; Takaya, Y.; Ogasawara, M. Org. Synth. 2002, 79, 84.

(37) Yamamoto, Y.; Takizawa, M.; Yu, X.; Miyaura, N. Angew. Chem., Int. Ed. 2008, 47, 928.

(38) Yu, X.; Yamamoto, Y.; Miyaura, N. Synlett 2009, 994.

(39) Yoshida, K.; Ogasawara, M.; Hayashi, T. J. Org. Chem. 2003, 68, 1901.

(40) Shintani, R.; Tsutsumi, Y.; Nagaosa, M.; Nishimura, T.; Hayashi, T. J. Am. Chem. Soc. 2009, 131, 13588.

(41) Negishi, E. Organometallics in Organic Synthesis; Wiley: New York, 1980.

(42) Hawner, C.; Müller, D.; Gremaud, L.; Felouat, A.; Woodward,

S.; Alexakis, A. Angew. Chem., Int. Ed. 2010, 49, 7769.

(43) Yoshida, K.; Hayashi, T. J. Am. Chem. Soc. 2003, 125, 2872.

(44) Hayashi, T.; Tokunaga, N.; Yoshida, K.; Han, J. W. J. Am. Chem. Soc. 2002, 124, 12102.

(45) Hayashi, T.; Kawai, M.; Tokunaga, N. Angew. Chem., Int. Ed. 2004, 43, 6125.

(46) Hayashi, T.; Tokunaga, N.; Inoue, K. Org. Lett. 2004, 6, 305.

(47) Kina, A.; Ueyama, K.; Hayashi, T. Org. Lett. 2005, 7, 5889.

(48) Tokunaga, N.; Hayashi, T. Tetrahedron: Asymmetry 2006, 17, 607.

(49) Shintani, R.; Tokunaga, N.; Doi, H.; Hayashi, T. J. Am. Chem. Soc. 2004, 126, 6240.

(50) Hayashi, T.; Yamamoto, S.; Tokunaga, N. Angew. Chem., Int. Ed. 2005, 44, 4224.

(51) Mori, A.; Danda, Y.; Fujii, T.; Hirabayashi, S.; Osakada, K. J. Am. Chem. Soc. 2001, 123, 10774.

(52) Huang, T.; Li, C. Chem. Commun. 2001, 2348.

(53) Murata, M.; Shimazaki, R.; Ishikura, M.; Watanabe, S.; Masuda, Y. Synthesis 2002, 717.

(54) Koike, T.; Du, X. L.; Mori, A.; Osakada, K. Synlett 2002, 301.

(55) Nakao, Y.; Chen, J.; Imanaka, H.; Hiyama, T.; Ichikawa, Y.;

Duan, W. L.; Shintani, R.; Hayashi, T. J. Am. Chem. Soc. 2007, 129, 9137.

(56) Miura, T.; Murakami, M. Chem. Commun. 2005, 5676.

(57) Venkatraman, S.; Meng, Y.; Li, C. J. Tetrahedron Lett. 2001, 42, 4459.

(58) Oi, S.; Moro, M.; Ono, S.; Inoue, Y. Chem. Lett. 1998, 83.

(59) Oi, S.; Moro, M.; Ito, H.; Honma, Y.; Miyano, S.; Inoue, Y. Tetrahedron 2002, 58, 91.

(60) Ding, R.; Chen, Y. J.; Wang, D.; Li, C. J. Synlett 2001, 1470.

(61) Venkatraman, S.; Li, C. J. Tetrahedron Lett. 2001, 42, 781.

(62) Huang, T. S.; Venkatraman, S.; Meng, Y.; Nguyen, T. V.; Kort,

D.; Wang, D.; Ding, R.; Li, C. J. Pure Appl. Chem. 2001, 73, 1315.

(63) Nishimura, T.; Katoh, T.; Hayashi, T. Angew. Chem., Int. Ed. 2007, 46, 4937.

(64) Sakuma, S.; Miyaura, N. J. Org. Chem. 2001, 66, 8944.

(65) Lukin, K.; Zhang, Q. Y.; Leanna, M. R. J. Org. Chem. 2009, 74, 929.

(66) Oi, S.; Moro, M.; Inoue, Y. Chem. Commun. 1997, 1621.

(67) Oi, S.; Moro, M.; Ono, S.; Inoue, Y. Organometallics 2001, 20, 1036.

(68) Oi, S.; Moro, M.; Ono, S.; Inoue, Y. Org. Lett. 2002, 4, 667.

(69) Oi, S.; Taira, A.; Honma, Y.; Inoue, Y. Org. Lett. 2003, 5, 97.

(70) Oi, S.; Sato, T.; Inoue, Y. Tetrahedron Lett. 2004, 45, 5051.

(71) Oi, S.; Taira, A.; Honma, Y.; Sato, T.; Inoue, Y. Tetrahedron: Asymmetry 2006, 17, 598.

(72) Itoh, T.; Mase, T.; Nishikata, T.; Iyama, T.; Tachikawa, H.;

Kobayashi, Y.; Yarnamoto, Y.; Miyaura, N. Tetrahedron 2006, 62, 9610. (73) Sakuma, S.; Sakai, M.; Itooka, R.; Miyaura, N. J. Org. Chem.

2000, 65, 5951. (74) Yamamoto, Y.; Kurihara, K.; Sugishita, N.; Oshita, K.; Piao,

D. G.; Miyaura, N. Chem. Lett. 2005, 34, 1224.

(75) Shimada, T.; Suda, M.; Nagano, T.; Kakiuchi, K. J. Org. Chem. 2005, 70, 10178.

(76) Amengual, R.; Michelet, V.; Genêt, J. -P. Synlett 2002, 1791.

(77) Reetz, M. T.; Moulin, D.; Gosberg, A. Org. Lett. 2001, 3, 4083.

(78) Kurihara, K.; Sugishita, N.; Oshita, K.; Piao, D.; Yamamoto, Y.; Miyaura, N. J. Organomet. Chem. 2007, 692, 428.

(79) Vandyck, K.; Matthys, B.; Willen, M.; Robeyns, K.; Van Meervelt, L.; Van der Eycken, J. Org. Lett. 2006, 8, 363.

(80) Gök, Y.; Noël, T.; Van der Eycken, J. Tetrahedron: Asymmetry 2010, 21, 2768.

(81) Shi, Q.; Xu, L. J.; Li, X. S.; Jia, X.; Wang, R. H.; Au-Yeung, T. T. L.; Chan, A. S. C.; Hayashi, T.; Cao, R.; Hong, M. C. Tetrahedron Lett. 2003, 44, 6505.

(82) Jeulin, S.; Duprat de Paule, S.; Ratovelomanana-Vidal, V.; Genêt, J. -P.; Champion, N.; Dellis, P. Angew. Chem., Int. Ed. 2004, 43, 320.

(83) Mashima, K.; Kusano, K.; Saito, N.; Matsumura, Y.; Nozaki, K.; Kumobayashi, H.; Sayo, N.; Hori, Y.; Ishizaki, Akutagawa, S.; Takaya, H. J. Org. Chem. 1994, 59, 3064.

(84) Korenaga, T.; Osaki, K.; Maenishi, R.; Sakai, T. Org. Lett. 2009, 11, 2325.

- <span id="page-22-0"></span>(85) Korenaga, T.; Maenishi, R.; Hayashi, K.; Sakai, T. Adv. Synth. Catal. 2010, 352, 3247.
- (86) Korenaga, K.; Maenishi, R.; Osaki, K.; Sakai, T. Heterocycles 2010, 80, 157.
- (87) Berhal, F.; Esseiva, O.; Martin, C. -H.; Tone, H.; Genêt, J. -P.; Ayad, T.; Ratovelomanana-Vidal, V. Org. Lett. 2011, 13, 2806.
- (88) Berhal, F.; Wu, Z.; Genêt, J. -P.; Ayad, T.; Ratovelomanana-Vidal, V. J. Org. Chem. 2011, 76, 6320.
- (89) Madec, J.; Michaud, G.; Genêt, J. P.; Marinetti, A. Tetrahedron: Asymmetry 2004, 15, 2253.
- (90) Otomaru, Y.; Senda, T.; Hayashi, T. Org. Lett. 2004, 6, 3357.
- (91) Stemmler, R. T.; Bolm, C. J. Org. Chem. 2005, 70, 9925.
- (92) Kromm, K.; Eichenseher, S.; Prommesberger, M.; Hampel, F.; Gladysz, J. A. Eur. J. Inorg. Chem. 2005, 2983.
- (93) Takaya, Y.; Ogasawara, M.; Hayashi, T. Chirality 2000, 12, 469. (94) Imamoto, T.; Sugita, K.; Yoshida, K. J. Am. Chem. Soc. 2005, 127, 11934.
- (95) Imamoto, T.; Saitoh, Y.; Koide, A.; Ogura, T.; Yoshida, K. Angew. Chem., Int. Ed. 2007, 46, 8636.
- (96) van den Berg, M.; Minnaard, A. J.; Schudde, E. P.; van Esch, J.; de Vries, A. H. M.; de Vries, J. G.; Feringa, B. L. J. Am. Chem. Soc.
- 2000, 122, 11539. (97) Martina, S. L. X.; Minnaard, A. J.; Hessen, B.; Feringa, B. L.
- Tetrahedron Lett. 2005, 46, 7159.
- (98) Duursma, A.; Boiteau, J. G.; Lefort, L.; Boogers, J. A.; de Vries, A. H.; De Vries, J.; Minnaard, A. J.; Feringa, B. L. J. Org. Chem. 2004,
- 69, 8045.
- (99) Boiteau, J. G.; Imbos, F.; Minnaard, A. J.; Feringa, B. L. Org. Lett. 2003, 5, 681.
- (100) Boiteau, J. G.; Minnaard, A. J.; Feringa, B. L. J. Org. Chem. 2003, 68, 9481.
- (101) Duursma, A.; Hoen, R.; Schuppan, J.; Hulst, R.; Minnaard, A. J.; Feringa, B. L. Org. Lett. 2003, 5, 3111.
- (102) Jagt, R. B. C.; de Vries, J. G.; Feringa, B. L.; Minnaard, A. J. Org. Lett. 2005, 7, 2433.
- (103) Jagt, R. B. C.; Toullec, P. Y.; Schudde, E. P.; De Vries, J. G.; Feringa, B. L.; Minnaard, A. J. J. Comb. Chem. 2007, 9, 407.
- (104) Iuliano, A.; Facchetti, S.; Funaioli, T. Chem. Commun. 2009, 457.
- (105) Monti, C.; Gennari, C.; Piarulli, U. Chem. Commun. 2005, 5281.
- (106) Ma, Y. D.; Song, C.; Ma, C. Q.; Sun, Z. J.; Chai, Q.; Andrus, M. B. Angew. Chem., Int. Ed. 2003, 42, 5871.
- (107) Facchetti, S.; Cavallini, I.; Funaioli, T.; Marchetti, F.; Iuliano, A. Organometallics 2009, 28, 4150.
- (108) Hayashi, T.; Ueyama, K.; Tokunaga, N.; Yoshida, H. J. Am. Chem. Soc. 2003, 125, 11508.
- (109) Defieber, C.; Paquin, J.; Serna, S.; Carreira, E. M. Org. Lett. 2004, 6, 3873.
- (110) Paquin, J.; Stephenson, C. R. J.; Defieber, C.; Carreira, E. M. Org. Lett. 2005, 7, 3821.
- (111) Otomaru, Y.; Okamoto, K.; Shintani, R.; Hayashi, T. J. Org. Chem. 2005, 70, 2503.
- (112) Lang, F.; Breher, F.; Stein, D.; Grutzmacher, H. Organometallics 2005, 24, 2997.
- (113) Helbig, S.; Sauer, S.; Cramer, N.; Laschat, S.; Baro, A.; Frey, W. Adv. Synth. Catal. 2007, 349, 2331.
- (114) Nishimura, T.; Nagaosa, M.; Hayashi, T. Chem. Lett. 2008, 37, 860.
- (115) Okamoto, K.; Hayashi, T.; Rawal, V. H. Org. Lett. 2008, 10, 4387.
- (116) Gendrineau, T.; Chuzel, O.; Eijsberg, H.; Genêt, J. P.; Darses, S. Angew. Chem., Int. Ed. 2008, 47, 7669.
- (117) Shintani, R.; Ichikawa, Y.; Takatsu, K.; Chen, F. -X.; Hayashi, T. J. Org. Chem. 2009, 74, 869.
- (118) Fischer, C.; Defieber, C.; Takeyuki, S.; Carreira, E. M. J. Am. Chem. Soc. 2004, 126, 1628.
- (119) Berthon-Gelloz, G.; Hayashi, T. J. Org. Chem. 2006, 71, 8957.
- (120) Noël, T.; Vandyck, K.; Van der Eycken, J. Tetrahedron 2007, 63, 12961.
- (121) Brown, M. K.; Corey, E. J. Org. Lett. 2010, 12, 172.
- (122) Hayashi, T.; Tokunaga, N.; Okamoto, K.; Shintani, R. Chem. Lett. 2005, 34, 1480.
- (123) Shintani, R.; Kimura, T.; Hayashi, T. Chem. Commun. 2005, 3213.
- (124) Shintani, R.; Okamoto, K.; Hayashi, T. Org. Lett. 2005, 7, 4757. (125) Paquin, J. F.; Defieber, C.; Stephenson, C. R. J.; Carreira, E. M.
- J. Am. Chem. Soc. 2005, 127, 10850.
- (126) Chen, F.; Kina, A.; Hayashi, T. Org. Lett. 2006, 8, 341.
- (127) Tokunaga, N.; Hayashi, T. Adv. Synth. Catal. 2007, 349, 513. (128) Soergel, S.; Tokunaga, N.; Sasaki, K.; Okamoto, K.; Hayashi, T.
- Org. Lett. 2008, 10, 589. (129) Okamoto, K.; Hayashi, T.; Rawal, V. H. Chem. Commun. 2009,
- 4815.
- (130) Otomaru, Y.; Tokunaga, N.; Shintani, R.; Hayashi, T. Org. Lett. 2005, 7, 307.
- (131) Otomaru, Y.; Kina, A.; Shintani, R.; Hayashi, T. Tetrahedron: Asymmetry 2005, 16, 1673.
- (132) Wang, Z. -Q.; Feng, C. -G.; Xu, M. -H.; Lin, G. -Q. J. Am. Chem. Soc. 2007, 129, 5336.
- (133) Wang, Y. -Z.; Hu, X. -C.; Du, H. -F. Org. Lett. 2010, 12, 5482. (134) Li, Q.; Dong, Z.; Yu, Z. -X. Org. Lett. 2011, 13, 1122.
- (135) Chen, M. S.; White, C. M. J. Am. Chem. Soc. 2004, 126, 1346.
- (136) Chen, M. S.; Prabagaran, N.; Labenz, N. A.; White, C. M. J. Am. Chem. Soc. 2005, 127, 6970.
- (137) Mariz, R.; Luan, X.; Gatti, M.; Linden, A.; Dorta, R. J. Am. Chem. Soc. 2008, 130, 2172.
- (138) Bü rgi, J. J.; Mariz, R.; Gatti, M.; Drinkel, E.; Luan, X.; Blumentritt, S.; Linden, A.; Dorta, R. Angew. Chem., Int. Ed. 2009, 48, 2768.
- (139) Tokunaga, N.; Otomaru, Y.; Okamoto, K.; Ueyama, K.; Shintani, R.; Hayashi, T. J. Am. Chem. Soc. 2004, 126, 13584.
- (140) Chen, Q.-A.; Dong, X.; Chen, M.-W; Wang, D.-S.; Zhou, Y.-G.; Li, Y.-X. Org. Lett. 2010, 12, 1928.
- (141) Chen, J.; Chen, J.-M.; Lang, F.; Zhang, X.-Y.; Cun, L.-F.; Zhu, J.; Deng, J.-G.; Liao, J. J. Am. Chem. Soc. 2010, 132, 4552.
- (142) Han, F.-Z.; Chen, G.-H.; Zhang, X.-Y.; Liao, J. Eur. J. Org. Chem. 2011, 2928.
- (143) Zhang, X.-Y.; Chen, J.; Han, F.-Z.; Cun, L.-F.; Liao, J. Eur. J. Org. Chem. 2011, 1443.
- (144) Korenaga, T.; Hayashi, K.; Akaki, Y.; Maenishi, R.; Sakai, T. Org. Lett. 2011, 13, 2022.
- (145) Shintani, R.; Duan, W.-L.; Nagano, T.; Okada, A.; Hayashi, T. Angew. Chem., Int. Ed. 2005, 44, 4611.
- (146) Duan, W.-L.; Iwamura, H.; Shintani, R.; Hayashi, T. J. Am. Chem. Soc. 2007, 129, 2130.
- (147) Piras, E.; Lang, F.; Ruegger, H.; Stein, D.; Worle, M.; Grützmacher, H. Chem.-Eur. J. 2006, 12, 5849.
- (148) Kasak, P.; Arion, V. B.; Widhalm, M. Tetrahedron: Asymmetry 2006, 17, 3084.
- (149) Defieber, C.; Ariger, M. A.; Moriel, P.; Carreira, E. M. Angew. Chem., Int. Ed. 2007, 46, 3139.
- (150) Mariz, R.; Briceno, A.; Dorta, R. Organometallics 2008, 27, 6605.
- (151) Minuth, T.; Boysen, M. M. L. Org. Lett. 2009, 11, 4212.
- (152) Grugel, H.; Minuth, T.; Boyse, M. M. K. Synthesis 2010, 3248.
- (153) Kuriyama, M.; Tomioka, K. Tetrahedron Lett. 2001, 42, 921.
- (154) Kuriyama, M.; Nagai, K.; Yamada, K.; Miwa, Y.; Taga, T.; Tomioka, K. J. Am. Chem. Soc. 2002, 124, 8932.
- (155) Chen, Q.; Soeta, T.; Kuriyama, M.; Yamada, K. I.; Tomioka, K. Adv. Synth. Catal. 2006, 348, 2604.
- (156) Lang, F.; Li, D.; Chen, J.-M.; Chen, J.; Li, L.-C.; Cun, L.-F.; Zhu, J.; Deng, J.-G.; Liao, J. Adv. Synth. Catal. 2010, 352, 843.
- (157) Han, F.-Z.; Chen, J.; Zhang, X.-Y.; Liu, J.-B.; Cun, L.-F.; Zhu, J.; Deng, J.-G.; Liao, J. Tetrahedron Lett. 2011, 52, 830.
- (158) Becht, J.-M.; Bappert, E.; Helmchen, G. Adv. Synth. Catal. 2005, 347, 1495.
- <span id="page-23-0"></span>(159) Hahn, B. T.; Tewes, F.; Frö hlich, R.; Glorius, F. Angew. Chem., Int. Ed. 2010, 49, 1143.
- (160) Kuuloja, N.; Tois, J.; Franzén, R. Tetrahedron: Asymmetry 2011, 22, 468.
- (161) Thaler, T.; Guo, L.-N.; Steib, A. K.; Raducan, M.; Karaghiosoff, K.; Mayer, P.; Knochel, P. Org. Lett. 2011, 13, 3182.
- (162) Chen, G.-H.; Gui, J.-Y.; Li, L.-C.; Liao, J. Angew. Chem., Int. Ed. 2011, 50, 7681.
- (163) Xue, F.; Li, X.-C.; Wan, B.-S. J. Org. Chem. 2011, 76, 7256.
- (164) Qi, W.-Y.; Zhu, T.-S.; Xu, M.-H. Org. Lett. 2011, 13, 3410.
- (165) Jin, S.-S.; Wang, H.; Xu, M.-H. Chem. Commun. 2011, 7230.
- (166) Sun, X.-W.; Xu, M.-H.; Lin, G.-Q. Org. Lett. 2006, 8, 4979.
- (167) Sun, X.-W.; Liu, M.; Xu, M.-H.; Lin, G.-Q. Org. Lett. 2008, 10, 1259.
- (168) Liu, M.; Sun, X.-W.; Xu, M.-H.; Lin, G.-Q. Chem.—Eur. J. 2009, 15, 10217.
- (169) Liu, M.; Shen, A.; Sun, X.-W.; Deng, F.; Xu, M.-H.; Lin, G.-Q. Chem. Commun. 2010, 8460.
- (170) Feng, X.-Q.; Wang, Y.-Z.; Wei, B.-B.; Yang, J.; Du, H.-F. Org. Lett. 2011, 13, 3300.
- (171) Feng, X.-Q.; Wei, B.-B.; Yang, J.; Du, H.-F. Org. Biomol. Chem. 2011, 9, 5927.
- (172) Ueda, M.; Miyaura, N. J. Org. Chem. 2000, 65, 4450.
- (173) Brock, S.; Hose, D. R. J.; Moseley, J. D.; Parker, A. J.; Patel, I.; Williams, A. J. Org. Process Res. Dev. 2008, 12, 496.
- (174) Shintani, R.; Hayashi, T. Org. Lett. 2011, 13, 350.
- (175) Paquin, J.-F.; Stephenson, C. R. J.; Defieber, C.; Carreira, E. M. Org. Lett. 2005, 7, 3821.
- (176) Chen, G.; Tokunaga, N.; Hayashi, T. Org. Lett. 2005, 7, 2285.
- (177) Meyer, O.; Becht, J. M.; Helmchen, G. Synlett 2003, 1539.
- (178) Becht, J.-M.; Meyer, O.; Helmchen, G. Synthesis 2003, 2805.
- (179) Nishimura, T.; Wang, J.; Nagaosa, M.; Okamoto, Z.; Shintani,
- R.; Kwong, F.-Y.; Yu, W.-Y.; Chan, A. S. C.; Hayashi, T. J. Am. Chem. Soc. 2010, 132, 464.
- (180) Zoute, L.; Kociok-Kö hn, G.; Frost, C. G. Org. Lett. 2009, 11, 2491.
- (181) Hellal, M.; Cuny, G. D. Org. Lett. 2010, 12, 4628.
- (182) Shintani, R.; Ueyama, K.; Yamada, I.; Hayashi, T. Org. Lett. 2004, 6, 3425.
- (183) Pucheault, M.; Michaut, V.; Darses, S.; Genêt, J.-P. Tetrahedron Lett. 2004, 45, 4729.
- (184) Shao, C.; Yu, H.-J.; Wu, N.-Y.; Tian, P.; Wang, R.; Feng, C.-G.; Lin, G.-Q. Org. Lett. 2011, 13, 788.
- (185) Shintani, R.; Duan, W. L.; Hayashi, T. J. Am. Chem. Soc. 2006, 128, 5628.
- (186) Duan, W. L.; Imazaki, Y.; Shintani, R.; Hayashi, T. Tetrahedron 2007, 63, 8529.
- (187) Hayashi, T.; Senda, T.; Ogasawara, M. J. Am. Chem. Soc. 2000, 122, 10716.
- (188) Burgey, C. S.; Paone, D. V.; Shaw, A. W.; Deng, J. Z.; Nguyen,
- D. N.; Potteiger, C. M.; Graham, S. L.; Vacca, J. P.; Williams, T. M. Org. Lett. 2008, 10, 3235.
- (189) Wang, Z.-Q.; Feng, C.-G.; Zhang, S.-S.; Xu, M.-H.; Lin, G.-Q. Angew. Chem., Int. Ed. 2010, 49, 5780.
- (190) Hayashi, T.; Senda, T.; Takaya, Y.; Ogasawara, M. J. Am. Chem. Soc. 1999, 121, 11591.
- (191) Mauleon, P.; Carretero, J. C. Org. Lett. 2004, 6, 3195.
- (192) Mauleon, P.; Alonso, I.; Rivero, M. R.; Carretero, J. C. J. Org. Chem. 2007, 72, 9924.
- (193) Mauleón, P.; Carretero, J. Chem. Commun. 2005, 4961.
- (194) Plesniak, K.; Zarecki, A.; Wicha, J. Top. Curr. Chem. 2007, 275, 163.
- (195) Murakami, M.; Igawa, H. Chem. Commun. 2002, 390.
- (196) Oguma, K.; Miura, M.; Satoh, T.; Nomura, M. J. Am. Chem. Soc. 2000, 122, 10464.
- (197) Lautens, M.; Dockendorff, C.; Fagnou, K.; Malicki, A. Org. Lett. 2002, 4, 1311.
- (198) Menard, F.; Lautens, M. Angew. Chem., Int. Ed. 2008, 47, 2085.
- (199) Panteleev, J.; Menard, F.; Lautens, M. Adv. Synth. Catal. 2008, 350, 2893.
- (200) Bexrud, J.; Lautens, M. Org. Lett. 2010, 12, 3160.
- (201) Pattison, G.; Piraux, G.; Lam, H. W. J. Am. Chem. Soc. 2010, 132, 14373.
- (202) Saxena, A.; Lam, H. W. Chem. Sci 2011, 2, 2326.
- (203) Miura, T.; Takahashi, Y.; Murakami, M. Chem. Commun. 2007, 595.
- (204) Sasaki, K.; Hayashi, T. Angew. Chem., Int. Ed. 2010, 49, 8145.
- (205) de la Herrán, G.; Murcia, C.; Csákÿ, A. G. Org. Lett. 2005, 7, 5629.
- (206) Nishimura, T.; Makino, H.; Nagaosa, M.; Hayashi, T. J. Am. Chem. Soc. 2010, 132, 12865.
- (207) Krug, C.; Hartwig, J. F. J. Am. Chem. Soc. 2002, 124, 1674.
- (208) Boyd, S. E.; Field, L. D.; Hambley, T. W.; Partridge, M. G. Organometallics 1993, 12, 1720.
- (209) Jagt, R. B. C.; Toullec, P. Y.; de Vries, J. G.; Feringa, B. L.; Minnaard, A. J. Org. Biomol. Chem. 2006, 4, 773.
- (210) Duan, H.-F.; Xie, J.-H.; Shi, W.-J.; Zhang, Q.; Zhou, Q.-L. Org. Lett. 2006, 8, 1479.
- (211) Nishimura, T.; Kumamoto, H.; Nagaosa, M.; Hayashi, T. Chem. Commun. 2009, 5713.
- (212) Morikawa, S.; Michigami, K.; Amii, H. Org. Lett. 2010, 12, 2520.
- (213) Ma, Q.-S.; Ma, Y.-D.; Liu, X.; Duan, W.-Z.; Qu, B.; Song, C. Tetrahedron: Asymmetry 2010, 21, 292.
- (214) Xing, C.-H.; Liao, Y.-X.; He, P.; Hu, Q.-S. Chem. Commun. 2010, 3010.
- (215) Shintani, R.; Inoue, M.; Hayashi, T. Angew. Chem., Int. Ed. 2006, 45, 3353.
- (216) Martina, S. L. X.; Jagt, R. B. C.; de Vries, J. G.; Feringa, B. L.; Minnaard, A. J. Chem. Commun. 2006, 4093.
- (217) Duan, H.-F.; Xie, J.-H.; Qiao, X.-C.; Wang, L.-X.; Zhou, Q.-L. Angew. Chem., Int. Ed. 2008, 47, 4351.
- (218) Schmidt, F.; Stemmler, R. T.; Rudolph, J.; Bolm, C. Chem. Soc. Rev. 2006, 35, 454.
- (219) Weix, D. J.; Shi, Y.; Ellman, J. A. J. Am. Chem. Soc. 2005, 127, 1092.
- (220) Brak, K.; Ellman, J. A. J. Org. Chem. 2010, 75, 3147.
- (221) Bolshan, Y.; Batey, R. A. Org. Lett. 2005, 7, 1481.
- (222) Liu, G.; Cogan, D. A.; Owens, T. D.; Tang, T. P.; Ellman, J. A.
- J. Org. Chem. 1999, 64, 1278. (223) Davis, F. A.; McCoull, W. J. Org. Chem. 1999, 64, 3396.
- (224) Jayathilaka, L. P.; Deb, M.; Standaert, R. F. Org. Lett. 2004, 6,
- 3659. (225) Beenen, M. A.; Weix, D. J.; Ellman, J. A. J. Am. Chem. Soc.
- 2006, 128, 6304.
- (226) Truong, V. L.; Pfeiffer, J. Y. Tetrahedron Lett. 2009, 50, 1633.
- (227) Kuriyama, M.; Soeta, T.; Hao, X. Y.; Chen, O.; Tomioka, K.
- J. Am. Chem. Soc. 2004, 126, 8128.
- (228) Duan, H.-F.; Jia, Y.-X.; Wang, L.-X.; Zhou, Q.-L. Org. Lett. 2006, 8, 2567.
- (229) Wang, Z.-Q.; Feng, C.-G.; Xu, M.-H.; Lin, G.-Q. J. Am. Chem. Soc. 2007, 129, 5336.
- (230) Yang, H.-Y.; Xu, M.-H. Chem. Commun. 2010, 9223.
- (231) Shao, C.; Yu, H.-J.; Wu, N.-Y.; Feng, C.-G.; Lin, G.-Q. Org. Lett. 2010, 12, 3820.
- (232) Shintani, R.; Narui, R.; Tsutsumi, Y.; Hayashi, S.; Hayashi, T. Chem. Commun. 2011, 6123.
- (233) Cui, Z.; Yu, H.-J.; Yang, R.-F.; Gao, W.-Y.; Feng, C.-G.; Lin, G.-Q. J. Am. Chem. Soc. 2011, 133, 12394 ; see also ref 261.
- (234) Fukuyama, T.; Jow, C.-K.; Cheung, M. Tetrahedron Lett. 1995, 36, 6373.
- (235) Kan, T.; Fukuyama, T. Chem. Commun. 2004, 3[53.](#page-24-0)
- (236) Greene, T. W.; Wuts, P. G. M. Protective Groups in Organic Synthesis, 3rd ed.; Wiley: New York, 1999; p 609.
- (237) Jagt, R. B. C.; Toullec, P. Y.; Geerdink, D.; de Vries, J. G.; Feringa, B. L.; Minnaard, A. J. Angew. Chem., Int. Ed. 2006, 45, 2789. (238) Cao, Z.-P.; Du, H.-F. Org. Lett. 2010, 12, 2602.
- <span id="page-24-0"></span>(239) Crampton, R.; Woodward, S.; Fox, M. Adv. Synth. Catal. 2011, 353, 903.
- (240) Shintani, R.; Takeda, M.; Tsuji, T.; Hayashi, T. J. Am. Chem. Soc. 2010, 132, 13168.
- (241) Shintani, R.; Takeda, M.; Soh, Y.-T.; Ito, T.; Hayashi, T. Org. Lett. 2011, 13, 2977.
- (242) Hao, X.-Y.; Kuriyama, M.; Chen, Q.; Yamamoto, Y.; Yamada, K.-I.; Tomioka, K. Org. Lett. 2009, 11, 4470.
- (243) Hao, X.-Y.; Chen, Q.; Yamada, K.-I.; Yamamoto, Y.; Tomioka, K. Tetrahedron 2011, 67, 6469.
- (244) Shintani, R.; Soh, Y.-T.; Hayashi, T. Org. Lett. 2010, 12, 4106.
- (245) Nadeau, C.; Aly, S.; Belyk, K. J. Am. Chem. Soc. 2011, 133, 2878.
- (246) Trost, B. M. Science 1991, 254, 1471.
- (247) Guo, H. C.; Ma, J. A. Angew. Chem., Int. Ed. 2006, 45, 354.
- (248) Miura, T.; Murakami, M. Chem. Commun. 2007, 217.
- (249) Youn, S. W. Eur. J. Org. Chem. 2009, 2597.
- (250) Yoshida, K.; Ogasawara, M.; Hayashi, T. J. Am. Chem. Soc. 2002, 124, 10984.
- (251) Cauble, D. F.; Gipson, J. D.; Krische, M. J. J. Am. Chem. Soc. 2003, 125, 1110.
- (252) Bocknack, B. M.; Wang, L. C.; Krische, M. J. Proc. Natl. Acad. Sci. U.S.A. 2004, 101, 5421.
- (253) Miura, T.; Harumashi, T.; Murakami, M. Org. Lett. 2007, 9, 741.
- (254) Miura, T.; Murakami, M. Org. Lett. 2005, 7, 3339.
- (255) Shintani, R.; Okamoto, K.; Otomaru, Y.; Ueyama, K.; Hayashi, T. J. Am. Chem. Soc. 2005, 127, 54.
- (256) Shintani, R.; Okamoto, K.; Hayashi, T. Chem. Lett. 2005, 34, 1294.
- (257) Shintani, R.; Tsurusaki, A.; Okamoto, K.; Hayashi, T. Angew. Chem., Int. Ed. 2005, 44, 3909.
- (258) Shintani, R.; Isobe, S.; Takeda, M.; Hayashi, T. Angew. Chem., Int. Ed. 2010, 49, 3795.
- (259) Miura, T.; Shimada, M.; Murakami, M. Chem.-Asian J. 2006, 1, 868.
- (260) Miura, T.; Sasaki, T.; Nakazawa, H.; Murakami, M. J. Am. Chem. Soc. 2005, 127, 1390.
- (261) Wang, Z.-Q.; Xu, M.-H.; Lin, G.-Q. Synthesis 2010, 3263.